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- 211 Zinc induces distinct changes in the metabolism of reactive oxygen and nitrogen species
- 12 (ROS and RNS) in the roots of two *Brassica* species with different sensitivity to zinc
- 13 stress
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2 Running title: Zinc-induced changes in ROS and RNS metabolism of *Brassica* roots

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1 Abstract

- Background and Aims Zinc (Zn) is an essential micronutrient present naturally in soils but
- it can be accumulated in the environment by anthropogenic activities provoking different
- 4 types of damages to plants. Heavy metal stress stimulates the metabolism of reactive
- 5 oxygen and nitrogen species (ROS and RNS) in different plant species. This study assesses
- 6 the interplay of these two families of molecules to evaluate the putative nitro-oxidative
- stress response in roots of two *Brassica* species under high concentrations of zinc.
- 8 Methods Nine-day-old hydroponically grown Brassica juncea and Brassica napus
- 9 seedlings were treated with ZnSO₄ (0, 50, 150 and 300 μM) for seven days. Stress
- intensity was determined through the analyses of cell wall damages and cell viability. Key
- components of the metabolism of ROS and RNS including lipid peroxidation, enzymatic
- antioxidants, protein nitration and content of superoxide radical (O2⁻), nitric oxide (NO)
- and peroxynitrite (ONOO-) were also measured through biochemical and cellular
- 14 approaches.
- Key Results Under zinc stress, the analysis of morphological root damages and alterations
- of microelement homeostasis indicate that *B. juncea* is more tolerant to Zn stress than *B*.
- 17 *napus*. On the other hand, ROS and RNS parameters suggest that the oxidative components
- are predominant compared to the nitrosative ones in the root system of both species.
- Conclusions The results indicate a clear relationship between ROS and RNS metabolisms
- as mechanism of response against an excess of zinc stress which provokes a nitro-oxidative
- stress. However, the oxidative stress components seem to be more dominant than the
- 22 elements of the nitrosative stress in the root system of these two *Brassica* species.

1 Keywords

- 2 Brassica juncea, Brassica napus, excess zinc, reactive oxygen species, oxidative stress,
- 3 reactive nitrogen species, nitrosative stress, protein nitration

Introduction

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agriculture. While according to the World Health Organisation, 31% of the world's 4 5 population is potentially is at risk of zinc (Zn) deficiency (WHO 2005), the Zn contamination 6 also appears to be a growing problem over the last decades (Zarcinas et al. 2004). The most 7 important sources of zinc pollution in the environment are mostly anthropogenic, such as mining, waste disposal, electroplating or smelting (Bacon and Dinev 2005; Bi et al. 2006). 8 9 Being an essential micronutrient, zinc plays an important role as a cofactor in numerous 10 enzymes involved in protein synthesis and in carbohydrate, nucleic acid and lipid metabolism (Broadley et al. 2007). On the other hand, Zn excess may have a negative effect on plants. 11 Among others, seed germination and plant growth inhibition (Mrozek and Funicelli 1982, 12 Wang et al. 2009), changes in root development (Lingua et al. 2008), loss of membrane 13 integrity (Stoyanova and Doncheva 2002) or cell death (Chang et al. 2005) were determined 14 as the effects of zinc exposure. The mechanisms behind Zn toxicity are not completely 15 16 understood; competition for catalytic sites or transporters (González-Guerrero et al. 2005), 17 Zn-induced micronutrient-deficiency (Bonnet et al. 2000, Wang et al. 2009) or induction of oxidative stress (Wintz et al. 2003) were evidenced. 18 19 Non-redox active heavy metals such as zinc can cause oxidative stress by blocking essential functional groups in biomolecules because of their ability to bind strongly to oxygen, nitrogen 20 21 or sulphur atoms, hereby inactivating enzymes by binding to their cysteine residues (Nieboer 22 and Richardson 1980); or either able to replace another essential metal ions in their catalytic sites (Schützendübel and Polle 2002). During oxidative stress, reactive oxygen species (ROS), 23 such as superoxide anion (O2⁻), hydrogen peroxide (H2O2), and hydroxyl radicals (·OH) are 24

Heavy metal contamination is an increasingly serious problem for the environment and the

- commonly generated. High levels of ROS are able to damage macromolecules, thus the ROS
- 2 concentrations are needed to be strictly controlled by complex mechanisms in plants (Apel
- and Hirt, 2004). These include several enzymes such as ascorbate peroxidase (APX, EC
- 4 1.11.1.11), glutathione reductase (GR, EC 1.6.4.2), catalase (CAT, EC 1.11.1.6) superoxide
- 5 dismutase (SOD, EC 1.1.5.1.1), and non-enzymatic, soluble antioxidants such as glutathione
- 6 and ascorbate, among others.
- 7 Besides ROS, the term reactive nitrogen species (RNS) is extensively used to describe the 8 family of nitric oxide (NO)-related molecules, such as peroxynitrite (ONOO), dinitrogen trioxide (N₂O₃), dinitrogen tetraoxide (N₂O₄), S-nitrosoglutathione (GSNO), nitrogen dioxide 9 10 radical (NO₂), nitrosonium cation (NO⁺) and nitroxyl anion (NO⁻) (Wang et al. 2013). 11 Nitrosative stress as another stress process caused by environmental factors evolves as the consequence of RNS accumulation in the plant cells (Corpas et al 2007, 2011). However, the 12 two families of reactive molecules (ROS and RNS) are involved in overlapping signalling 13 processes; as a matter of fact the existence of nitro-oxidative stress has been reported to occur 14 under certain circumstances (Corpas and Barroso 2013). An excellent example for the ROS-15 16 RNS crosstalk is the reaction between O₂ and NO yielding ONOO, which is responsible for the protein tyrosine nitration becoming a good biomarker of nitrosative stress in plants 17 (Corpas et al 2007, 2013). Protein tyrosine nitration is a posttranslational modification 18 19 resulting in an addition of a nitro group (-NO₂) to one of the two equivalent ortho carbons in the aromatic ring of tyrosine residues (Gow et al 2004). It causes steric and electronic 20 perturbations, which modify the tyrosine's capability to function in electron transfer reactions 21 22 or to keep the proper protein conformation (van der Vliet et al 1999). Tyrosine nitration can affect the function of a protein in numerous ways: besides no effect on functions or function 23 gain, the most common result of tyrosine nitration is the inhibition of the protein's function 24 (Greenacre and Ischiropoulos 2001; Radi 2004). Furthermore, tyrosine nitration has the ability 25

- to influence several signal transduction pathways through the prevention of tyrosine
- 2 phosphorylation (Galetskiy et al 2011).
- 3 In most plants, the vacuoles of the root cells serve as the most important Zn storage thus
- 4 removing the metal from the root-shoot-leaf transport system and play a crucial role in basal
- 5 Zn tolerance (Arrivault et al. 2006). Further protection mechanisms, like cell wall alterations,
- 6 such as callose deposition are facilitating the survival of the plants by limiting the uptake and
- 7 translocation of heavy metals and by preventing the leakage of assimilates and other nutrients
- 8 (Sjölund 1997, Chen and Kim 2009). During callose deposition, the properties of the cell wall
- 9 are modified by adding extra layers of carbohydrates synthesised by callose synthase, a
- transmembrane protein in the outer plasma membrane (Kartusch 2003).
- Since heavy metals like zinc result in a massive loss of crop yield all over the world, the goal
- of this study was to investigate the morphological and physiological responses of two
- important crop plants, Indian mustard (Brassica juncea) and oilseed rape (Brassica napus) to
- 214 zinc excess. Furthermore, our aim was to determine the potential involvement of ROS and
- 15 RNS in the zinc sensitivity of *Brassica* species.

Materials and methods

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Plant material and growing conditions

- 4 Brassica juncea L. Czern. and Brassica napus L. seeds were surface-sterilized with 5% (v/v)
- 5 sodium hypochlorite and then placed onto perlite-filled Eppendorf tubes floating on full-
- 6 strength Hoagland solution. The nutrient solution contained 5 mM Ca(NO₃)₂, 5 mM KNO₃, 2
- 7 mM MgSO₄, 1 mM KH₂PO₄, 0.01 mM Fe-EDTA, 10 μM H₃BO₃, 1 μM MnSO₄, 5 μM
- 8 ZnSO₄, 0.5 μM CuSO₄, 0.1 μM (NH₄)₆Mo₇O₂₄ and 10 μM AlCl₃. Seedlings were
- 9 precultivated for nine days until the appearance of the first leaves and then the nutrient
- solution was changed and supplemented with 0 (control), 50, 150 and 300 μM ZnSO₄ for
- seven days. Control plants were grown in full strength Hoagland solution containing 5 µM
- $ZnSO_4$. The plants were kept in a greenhouse at a photon flux density of 150 μ mol m⁻² s⁻¹
- 13 (12/12h light/dark cycle) at a relative humidity of 55-60% and 25±2°C for seven days.
- 14 All chemicals used during the experiments were purchased from Sigma-Aldrich (St. Louis,
- MO, USA) unless stated otherwise.

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Element content analysis

- 18 The concentrations of microelements were measured by using inductively coupled plasma
- 19 mass spectrometry (ICP-MS, Thermo Scientific XSeries II, Asheville, USA) according to
- Lehotai et al. (2012). Root and shoot material of control, 50, 150 and 300 μ M Zn-treated B.
- 21 *juncea* and *B. napus* were harvested separately and rinsed with distilled water to remove the
- potentially attached Zn from their surface. After 72 h of drying at 70°C, 65% (w/v) nitric acid

- and 30% (w/v) hydrogen peroxide (both from Reanal, Budapest, Hungary) were added to the
- samples, which were subjected to 200°C and 1600W for 15 min. Values of Zn and other
- 3 microelement concentrations are given in $\mu g g^{-1}$ dry weight (DW).

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Morphological measurements

- 6 Fresh weights (g) of the root material were measured on the 7th day of the treatment using a
- 7 balance. The length of the primary root (cm) and the first six lateral roots from the root collar
- 8 (cm) were also determined manually. Also the visible lateral roots were counted and their
- 9 number is expressed as pieces/root. The root fresh weight and the primary root length are
- 10 expressed as % of control.

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Microscopic determination of zinc distribution, callose deposition, lipid peroxidation

and viability loss in the root tissues

For visualisation of Zn, root tips were equilibrated in PBS buffer (137 mM NaCl, 2.68 mM

KCl, 8.1 mM Na₂HPO₄, 1.47 mM KH₂PO₄, pH 7.4), and further incubated with 25 µM

Zinquin (ethyl (2-methyl-8-p-toluenesulphonamido-6-quinolyloxy)acetate) in PBS for 1 h at

room temperature in darkness according to Sarret et al. (2006). Callose deposition in the root

tissues was determined by image analysis using aniline blue according to Cao et al. (2011)

with slight modifications. Root samples were incubated in aniline blue solution (0.1%, w/v in

1M glycine) for 5 min, and then washed once with distilled water. Products of lipid

peroxidation (such as malondialdehydes) were visualized using Schiff's reagent, according to

Arasimowicz-Jelonek et al. (2009). Root tips were incubated in the dye solution for 20

minutes and then the reagent was replaced by 0.5% (w/v) K₂S₂O₅ (prepared in 0.05M HCl) for

- a further 20 minutes. For the determination of cell viability in the root tips, fluorescein
- 2 diacetate (FDA) staining was used according to Lehotai et al. (2011). Root segments were
- 3 incubated in 10 μM dye solution prepared in 10 mM MES (4-morpholineethanesulfonic acid)
- 4 buffer (pH 6.15) containing 50 mM KCl, then they were washed four times with MES/KCl.

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Detection of reactive oxygen- (ROS) and nitrogen species (RNS)

7 Dihydroethidium (DHE) was used for visualisation of superoxide anion contents in the root tips, which were incubated for 30 min in darkness at 37°C in 10 µM dye solution and were 8 9 washed twice with 10 mM Tris/HCl, pH 7.4 (Kolbert et al., 2012). For hydrogen peroxide detection, root segments were incubated in 50 µM AmplifluTM (10-acetyl-3,7-10 dihydroxyphenoxazine, ADHP or Amplex Red) solution and washed with 50 mM sodium 11 phosphate buffer, pH 7.5, according to Lehotai et al. (2012). The NO levels in Brassica root 12 tips were determined by 4-amino-5-methylamino-2',7'-difluorofluorescein diacetate (DAF-13 FM DA) (Kolbert et al., 2012). Root segments were incubated for 30 min in darkness at room 14 temperature in 10 µM dye solution, and were washed twice with 10 mM Tris/HCl buffer, pH 15 16 7.4. Although DAF-FM DA allows only semi-quantitative analysis, it is a reliable fluorophore 17 for in situ detection of NO in plant tissues, since it does not react with hydrogen peroxide or peroxynitrite, but it responds to NO donors and/or scavengers (Kolbert et al., 2012). For the in 18 situ and in vivo detection of peroxynitrite (ONOO⁻), 3'-(p-aminophenyl) fluorescein (APF) 19 20 was applied (Chaki et al., 2009). The ONOO - sensitivity of APF was proved in vitro, and it was also shown that the dye does not react with NO or H₂O₂ (Kolbert et al., 2012). Root 21 samples were incubated in darkness at room temperature in 10 µM dye solution for 1 h and 22 were washed twice with 10 mM Tris/HCl buffer, pH 7.4. 23

- 1 The roots of *Brassica* plants labelled with different fluorophores were investigated under a
- 2 Zeiss Axiovert 200M inverted microscope (Carl Zeiss, Jena, Germany) equipped with filter
- set 9 (exc.: 450-490 nm, em.: 515- ∞ nm) for DHE, filter set 10 (exc.: 450-490, em.: 515-565
- 4 nm) for APF, DAF-FM and FDA, filter set 20HE (exc.: 546/12, em.: 607/80) for Amplex
- 5 Red, or filter set 49 (exc.: 365 nm, em.: 445/50 nm) for aniline blue and Zinquin.
- 6 Fluorescence intensities (pixel intensity) in the meristematic zone of the primary roots were
- 7 measured on digital images using Axiovision Rel. 4.8 software within circles of 100 μm radii.

8 Visualization of intracellular zinc compartmentalization by confocal laser scanning

9 microscopy (CLSM)

- 10 Root samples were washed three times alternately in deionized water and in 10 mM
- 11 ethylenediaminetetraacetic acid (EDTA) before being incubated in 20 μM Zinpyr-1 solution
- 12 (in PBS) at room temperature in darkness for 3 h (Sinclair et al. 2007). Samples were rinsed in
- 13 deionized water, and immersed in 10 μM propidium iodide to label cell walls (Tsukagoshi et
- al. 2010). Samples were mounted in PBS and images were taken on a CLS microscope
- 15 (Olympus LSM 700, Olympus, Tokyo, Japan) using excitation at 488 nm with a 100mW Ar
- ion laser and a ×20 Plan Apo water immersion lens with fluorescein isothiocyanate (FITC)
- and PI filters. Images were processed with Olympus Fluoview FV100 software and were
- analysed using Fiji software (http://fiji.sc/Fiji, Schindelin et al., 2012).

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Measurement of the enzymatic antioxidant activity

- SOD (EC 1.15.1.1) activity was determined by measuring the ability of the enzyme to inhibit
- 22 the photochemical reduction of nitro blue tetrazolium (NBT) in the presence of riboflavin in
- 23 light (Dhindsa et al., 1981). For the enzyme extract, 250 mg plant material was grinded with

- 1 10 mg polyvinyl polypyrrolidone (PVPP) and 1 ml 50 mM phosphate buffer (pH 7.0, with 1
- 2 mM EDTA added). The enzyme activity is expressed in Unit \cdot g⁻¹ fresh weight; one unit (U)
- 3 of SOD corresponds to the amount of enzyme causing a 50% inhibition of NBT reduction in
- 4 light.
- 5 Activity of ascorbate peroxidase (APX; EC 1.11.1.11) was measured by monitoring the
- 6 decrease of ascorbate content at 265 nm (E=14 mM⁻¹ cm⁻¹) according to a modified method by
- 7 Nakano and Asada (1981). For the enzyme extract, 250 mg plant material was grinded with
- 8 1.5 ml extraction buffer containing 1mM EDTA, 50mM NaCl and 900 μM ascorbate. Data
- 9 are expressed as activity (Unit \cdot g⁻¹ fresh weight).

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SOD activity on native PAGE, isoform staining

- SOD isoforms were detected in gels by the modified method of Beauchamp and Fridovich
- 13 (1971). SOD isozymes were separated by non-denaturating PAGE on 10% acrylamide gels,
- 14 followed by incubating sequentially in 2.45 mM NBT for 20 min and in 28 μM riboflavin and
- 15 28 mM tetramethyl ethylene diamine (TEMED) for 15 min in darkness. Colourless SOD
- bands on a dark blue background were observed after light exposure. SOD isoforms were
- identified by incubating gels in 50 mM potassium phosphate buffer (pH 7.0) supplemented
- with 3 mM KCN (inhibits Cu/Zn SOD) or 5 mM H₂O₂ (inhibits both Cu/Zn- and Fe-SOD) for
- 30 min before staining with NBT. Mn-SODs are resistant to both inhibitors.

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Immunoprecipitation, SDS-PAGE and Western blotting

- 22 Crude extracts from plant material were immunoprecipitated by using Thermo Scientific
- 23 Pierce Crosslink Magnetic IP/Co-IP Kit (Hudson, NH, USA). The beads were cross-linked

with antibody against 3-nitrotyrosine. After purification, immunoprecipited samples were 1 subjected to sodium dodecyl sulphate-polyacrylamide gel electrophoresis (SDS-PAGE) on 2 12% acrylamide gels. For western blot analysis, proteins were transferred to PVDF 3 membranes using the wet blotting procedure. After transfer, membranes were used for cross-4 5 reactivity assays with rabbit polyclonal antibody against 3-nitrotyrosine diluted 1:2000 (Corpas et al. 2008). Immunodetection was performed by using affinity isolated goat anti-6 rabbit IgG-alkaline phosphatase secondary antibody in dilution of 1:10 000, and bands were 7 8 visualised by using NBT/BCIP reaction. As a positive control nitrated bovine serum albumin was used. 9

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Statistical analysis

All experiments were carried out at least two times. In each treatment at least 10-20 samples were measured. The results are expressed as mean \pm SE. Multiple comparison analyses were performed with SigmaStat 12 software using analysis of variance (ANOVA, P<0.05) and Duncan's test. In some cases, Microsoft Excel 2010 and Student's t-test were used (*P \leq 0.05, **P \leq 0.01, ***P \leq 0.001).

Results and discussion

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Zinc uptake, accumulation and microelement homeostasis in Brassica species

4 Increasing zinc concentrations in the nutrient solutions promoted significant increases of zinc 5 content of the root system in both species (Fig 1A). The two Brassica species showed no differences in their zinc uptake capacity; although in case of the highest zinc treatment B. 6 7 juncea accumulated slightly (~14%) more metal in its root system. Regarding to the shoots, 8 the treatments resulted in a concentration-dependent response of zinc content; however the values were lower by one order of magnitude than in the root. Moreover, the shoot system of 9 10 B. napus contained higher (~48, ~33 and ~14%, respectively) zinc levels compared to that of B. juncea (Fig 1B). Results suggest an efficient root-to-shoot zinc translocation in both 11 species; however B. napus showed a better transport capacity to the aerial parts in case of all 12 zinc treatments. Zinc is predominantly complexed with citric and malic acid in the xylem sap. 13 Moreover, small amounts of soluble Zn-phosphate were also found in the sap in case of 14 15 excess Zn (White et al. 1981). In our experimental system, the Brassica species at their early 16 developmental stage (16 days-old plants) proved to be zinc accumulators, since the transported amount of Zn is more than 0.1% of the shoot dry weight. Similar Zn accumulation 17 tendencies were found in the roots and shoots of 12-days-old Brassica ssp., which were 18 19 considered to be as moderate zinc accumulators with some potentiality for phytoremediation (Ebbs and Kochian 1997). 20

Besides zinc, the concentrations of other microelements (Cu, Mn, Fe) were also determined in the roots of *Brassica* species using the ICP-MS technique (Table 1). The applied Zn treatments resulted in elevated copper contents in the root system of both *Brassica* species

compared to the control. The lack of the zinc-copper antagonism can be explained by the discrimination between the divalent ions (Irving-Williams order). Zinc and copper use the same transporters, which can be up-regulated by Zn excess, although they prefer Cu more than Zn (Fraústo da Silva and Williams 2001). This can trigger the increase in Cu content in the two zinc-exposed Brassica species. Similarly to copper, iron contents of the root system showed a slight but significant increase in both species as the effect of zinc. In contrast, all the applied zinc concentrations led to the strong decrease of manganese content in the roots of both species. In Arabidopsis thaliana and Thlaspi caerulescens roots, zinc caused similar changes of iron and manganese contents (van de Mortel et al. 2006). Moreover, also in the roots of zinc-exposed Lolium perenne the manganese contents were significantly reduced (Monnet et al. 2001). The synergistic effect we found between iron and zinc suggests that both species may increase their iron uptake in order to avoid iron deficiency in leaves. In Arabidopsis roots, Zn excess notably induced the expression of the ferric-chelate reductase gene (FRO2), which contributed to the intensification of Fe uptake (van de Mortel et al. 2006). In the case of manganese, an antagonistic relationship with zinc seems to be the rule in the roots of both species.

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Tissue-specific and subcellular localization of zinc in the root of *Brassica* species

The tissue localization pattern of zinc in the root tips was visualized by Zinquin fluorophore. Homogenous, low-level Zn-dependent fluorescence was detected in the root tips of control plants. As the effect of the increasing external Zn concentration, the accumulation of the fluorescent signal was the most evident in the meristematic and transition zones (Fig 2), probably because of the greater permeability of the thin walls of meristem cells for zinc ions. In the cells of the elongation zone, lower fluorescence intensity was observed, while in the

treatments and also the root hairs showed zinc content (see Fig 2A). Similarly, in the root tips of *Solanum nigrum*, zinc exposition (400 µM ZnCl₂) caused the intensification of Zinquin fluorescence, but this fluorescent signal showed homogenous distribution within the tip (Xu et al. 2010). Although practically there were no difference between the zinc contents of the whole root system of the species (see Fig 1A), the Zinquin fluorescence of the meristematic zones proved to be higher in zinc-treated *B. juncea* (Fig 2B), which suggests the higher accumulation of zinc in those root tips compared to *B. napus*. The increment of the fluorescent signal proved to be concentration-dependent neither in *B. juncea* nor in *B. napus* (Fig 2B).

Subcellular zinc distribution within the root cells of zinc-treated *Brassica* was investigated by confocal microscopy in order to reveal the role of the cell wall in zinc binding. The zinc-dependent green fluorescence was the most intense in the walls of the epidermal cell layer of the root. Zinc localized also in the cytoplasm and/or in vacuoles and around the surface of the nuclei of these cells, which showed also PI-dependent fluorescence, suggesting that these cells are not viable. The localization of zinc in the root cell nuclei of zinc-exposed plants was shown also by Rathore et al. (1972). In the inner cell layers of the root, mainly the apoplast possessed zinc content, and most of the cells were alive (Fig 3). Similarly to the results of Küpper et al. (2000), root epidermal cells accumulate zinc mostly in their walls. The cell wall metabolic inactivity provides advantage for metal precipitation and exclusion from the cytoplasm (Krzesłowska 2011), which can ensure the survival in case of metal excess (Rout and Das 2003). Some early works suggested that zinc is associated with the carbohydrate components of the cell wall such as hemicelluloses and pectins (Diez-Altares and Bornemisza 1967, Turner and Marshall 1972). Recent studies revealed that low-methylesterified pectins are the most important metal-binding components of the cell wall (Krzesłowska 2011). The

- 1 epidermal cells suffer from cell death presumably because of the presence of zinc also in their
- 2 cytoplasm, while the inner cells (cortex) contain less zinc mainly in their walls and they
- 3 remain viable.

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Zinc-triggered changes in root architecture

With the increasing Zn concentrations, leaf area, fresh and dry weight of the shoot significantly decreased (data not shown) and chlorosis was also visible (Fig 4A). However, necrotic lesions on the leaf blades were not observed during the experimental period. Root system has a great importance during the life of the heavy metal-exposed plants, since it can contribute to tolerance e.g. by controlling metal uptake or storage of excess metal. These support the necessity of the detailed investigation of the root development, which revealed differences between the Brassica species. The root tip morphology was modified by zinc excess, since the meristematic and transition zones were narrower, while the diameter of the upper regions was visibly larger than in the control root (see Fig 2A). This zinc-induced morphological alteration was observed in both species, but it was more evident in case of B. napus. Moreover, root hair formation was remarkably induced by zinc excess especially in B. napus roots (see Fig 2A). Interestingly, the primary root (PR) elongation of B. juncea was not notably affected by zinc, while in case of B. napus it was significantly inhibited at all applied zinc concentrations (Fig 4B). Mild stress (50 µM ZnSO₄) resulted in a notable elevation of lateral root (LR) number in both species. Similarly, 150 µM ZnSO₄ increased the number of LRs, but the effect was much slighter in this case. Moreover, the highest applied Zn concentration did not affect the LR development of the species (Fig 4C). The length of the lateral roots was remarkably diminished by zinc exposure in both Brassica species; however the rate of the inhibition proved to be lower in 50 µM zinc-treated B. juncea than in B. napus (Fig 4D). Based on these data, the cell elongation and division processes in the primary and lateral roots are more sensitive to zinc excess than the anticlinal divisions of the pericycle cells during LR initiation. Indeed, the PR tips of zinc-treated sugarcane showed significantly reduced mitotic index and wide spectrum of cytotoxic effects (Jain et al. 2010). The fresh weight of the root system showed zinc-induced reduction only in B. napus, while in B. juncea zinc stress was not able to alter the fresh root biomass (Fig 4E). Results show that zinc excess modifies the root system architecture depending on its concentration and the effect was different in the species. Namely, mild zinc exposure (50 µM ZnSO₄) triggered the development of stress-induced morphogenetic response (SIMR) phenotype (Potters et al. 2009) only in B. napus, since it resulted in shorter PR and larger amount of (shorter) LRs. Similar stress-induced root development was observed e.g. in selenium or copper exposed Arabidopsis or chromium treated wheat plants (Lehotai et al. 2012, Pető et al. 2011, Hasnain et al. 1997). In contrast, in *B. juncea* LR formation was induced (more significantly than in *B*. napus) and the PR elongation was not affected by zinc stress, which led to the development of an extended root system compared to the control plants. It can be assumed that these developmental changes can be parts of the acclimation process, because they can ensure better nutrient and water uptake thus survival of the *B. juncea* plant.

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Zinc stress provokes changes in the cell wall structure

Cell wall alterations, such as lignification or callose deposition, can help the plant cells tolerating excess heavy metal (HM) by serving as a physical barrier, thus preventing the heavy metals entering the cytoplasm. Besides their role in HM tolerance these cell wall modifications can partly be responsible for growth diminution as well.

Under copper stress a hydrogen peroxide-dependent lignin formation in the lateral roots of both *Brassica* species was found (Feigl et al. 2013), but zinc-induced lignification was not detectable in the root system (data not shown). On the other hand, the results show that excess Zn caused significant callose deposition in the roots of both species and this callose content increment was more pronounced in *B. napus* (Fig 5). Similarly, callose accumulation was observed in zinc-treated bean plants (Peterson and Rauser 1979). The deposited callose could inhibit root growth by decreasing cell wall loosening, thus preventing the passage of signal molecules or inhibit the symplastic supply of carbon required for root growth (Jones et al. 2006, Piršelová et al. 2012). In comparison, there was not significant callose deposition induced by copper stress either in *B. juncea* or in *B. napus* (Feigl et al. 2013), so hereby we can state that this cell wall modification is a heavy metal-dependent process, but it is independent from the plant species.

The Brassica species show different sensitivity to zinc stress

We characterized the degree of zinc sensitivity by detecting the viability of the root meristem (fluorescein diacetate labelling) and calculating the tolerance index (%) based on PR elongation. The root meristem cells of *B. juncea* remained fully viable even in case of 300 μM ZnSO₄ treatment, while root meristem of *B. napus* underwent significant viability loss as the effect of zinc exposure (Fig 6). Based on the results, the viability status of the primary root meristem cells are in accordance with the elongation capability of the root (see Fig 4B). The tolerance indexes of zinc-treated *B. juncea* showed no decrease at higher concentrations of external Zn (Control: 100%, 50 μM ZnSO₄: 97%, 150 μM ZnSO₄: 119% and 300 μM ZnSO₄: 107%); however they significantly decreased in case of zinc-exposed *B. napus* (Control: 100%, 50 μM ZnSO₄: 61%, 150 μM ZnSO₄: 50% and 300 μM ZnSO₄: 52%). The results

- show that B. juncea possesses remarkable zinc tolerance compared to B. napus, which
- 2 supports the species specificity of zinc sensitivity.

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Altered metabolism of ROS and RNS in Zn-exposed Brassica species

5 The effect of zinc excess on the ROS, RNS and antioxidants levels was accomplished in this 6 work. In the roots of B. juncea, the level of superoxide anion significantly decreased as a 7 consequence of zinc excess (Fig 7A), which can be explained by the enhancement of SOD 8 activity (Fig 7B). In contrast, superoxide levels of 50 and 150 µM Zn-treated B. napus roots showed a significant increment, which was accompanied by the increased SOD activity. 9 10 These suggest that the elevated SOD activity was not able to compensate the formation of superoxide anion in case of 50 and 150 μM Zn; although it could reduce the superoxide 11 content during severe Zn stress (300 µM). We separated the different SOD isoforms by native 12 PAGE and five activity bands were identified in case of both species (Fig 8). The obtained 13 pattern is in agreement with the result published by Cohu and Pilon (2007) in case of B. 14 15 juncea, however we found a different configuration of SOD isoforms in B. napus than it was 16 published by Abedi and Paniyat (2010). The experiments with specific inhibitors showed that the uppermost band represented a Mn-SOD isoform, which activity decreased by the 17 increasing Zn concentrations in both species, but especially in B. napus. The diminution of 18 19 Mn-SOD activity can be explained by the reduced availability of manganese as previously showed in Table 1. The Fe-SOD isoform was only hardly visible in case of *B. juncea* and only 20 21 present in the control sample of B. napus. The last three bands showed Cu/Zn-SODs, which 22 strengths were in correlation with the overall SOD activity (see Fig 6B). Similarly to our results, the decrease of the activity of all three isoenzymes was published e.g. in cadmium-23 exposed pea plants (Sandalio et al. 2001). 24

- 1 The level of H₂O₂ remained low in zinc-treated *B. juncea* (Fig 7C), and the pattern of APX
- 2 activity (Fig 7D) could partly explain the hydrogen peroxide profile. On the contrary, in B.
- 3 napus the highest applied Zn concentration resulted in an extreme H₂O₂ accumulation, but
- 4 APX did not vary comparing to control plants (Fig 7C and D, respectively). As it has been
- 5 reported earlier, zinc-triggered ROS formation and modification of antioxidant capacity was
- 6 published e.g. in sugarcane, bean, maize or pea (Jain et al. 2010, Chaoui et al. 1997, Lozano-
- 7 Rodríguez et al. 1997).
- 8 In the root tips of both examined species, NO formation was detectable in a concentrationdependent manner; however this elevation was statistically significant only in the roots of B. 9 10 juncea (Fig 7E). Several possible mechanisms of NO formation can exist in this system. Xu et 11 al. (2010) published that zinc-induced Fe-deficiency can be partially responsible for NO production in Solanum nigrum root tips, although in our experiments, zinc-induced Fe 12 deficiency was not observed (see Table 1). The major enzymatic source of NO in the roots is 13 nitrate reductase, but this activity was not influenced by zinc excess in Brassica roots (Bartha 14 et al. 2005). Furthermore, the transition metal-triggered decomposition of NO pools such as 15 16 S-nitrosoglutathione (Smith and Dasgupta 2000) may result in NO liberation in Zn-exposed Brassica roots, but this possibility remains to be elucidated. Nitric oxide may react with 17 superoxide anion yielding peroxynitrite (ONOO⁻), a powerful oxidative and nitrosative agent 18 (Arasimowicz-Jelonek and Floryszak-Wieczorek 2011). The significant zinc-induced 19 enhancement of peroxynitrite content in both species (Fig 7F) may explain the moderate NO 20 accumulation, since part of the formed NO possibly transformed into peroxynitrite. This 21 22 hypothesis can be supported by the decreasing superoxide levels in B. juncea (Fig 7B); while in B. napus superoxide levels remained high (Fig 7B) and less peroxynitrite is perhaps being 23 produced through this pathway (Fig 7F). The SOD system is possibly playing an important 24 25 role in the regulation of the peroxynitrite formation, by modulating the levels of superoxide

- 1 radicals driven to this reaction. The representative fluorescent microscopic images of the root
- tips stained with different fluorophores can be seen in Fig 7G.
- The significant and zinc concentration-dependent peroxynitrite formation in both species predicted protein tyrosine nitration and, therefore, this event was studied by Western blot
- 5 analysis using an antibody against nitro-tyrosine (Fig 9). The presence of seven nitrotyrosine-
- 6 immunopositive protein bands in the untreated samples suggests that a part of the protein pool
- 7 is nitrated even under control circumstances. Similarly, a basal nitration state of proteins was
- 8 published in different plant species such as sunflower, Citrus and pea (see Chaki et al., 2009;
- 9 Begara-Morales et al 2013; Corpas et al. 2013). We observed strengthening of the same seven
- 10 protein bands due to the effect of 300 µM ZnSO₄, which suggests the intensification of
- protein nitration induced by zinc excess. The enhancement of nitration levels was pronounced
- in both species, which implies that the proteome of both species are sensitive to nitrosative
- 13 modification. Similarly, intensified tyrosine nitration was observed in salt-stressed olive
- leaves as well as in leaves of cold-treated pea or in water-stressed *Lotus japonicus* (Corpas et
- al. 2008, Valderrama et al. 2006 Signorelli et al. 2013); however to our knowledge this is the
- 16 first which demonstrates heavy metal-induced protein nitration.
- Peroxynitrite through the formation of peroxynitrous acid (ONOOH) can lead to lipid
- 18 peroxidation, which product malondialdehyde (MDA) can be detected in situ by
- 19 histochemically (Arasimowicz-Jelonek et al. 2009). During the microscopic investigation
- 20 using the Schiff's staining procedure, the root tips of *B. napus* showed slight but visible pink
- 21 colorization reflecting the zinc-induced increment of the MDA content (Fig 10). In contrast,
- zinc-treated B. juncea root tips remained unstained. These suggest that the root tip cells of B.
- 23 napus suffered oxidative membrane damage, while in the root tips of B. juncea there was no
- 24 detectable lipid peroxidation.

Conclusions

Taken together, these results clearly show that the morphological and physiological responses 3 of Brassica species to zinc stress are different. In the roots B. juncea, possessing better zinc 4 5 resistance, only a slight ROS formation, activation of antioxidant enzymes (SOD, APX) and 6 no remarkable lipid peroxidation were observed, which reflect the lack of a zinc-induced serious oxidative stress. Although, the significant production of RNS (NO and ONOO-) and 7 the occurrence of protein nitration reveal a zinc-triggered secondary, nitrosative stress in B. 8 9 juncea. Contrary, as the effect of zinc exposure, nitro-oxidative stress occurred in the more 10 sensitive B. napus as a consequence of ROS and RNS accumulation, lipid peroxidation and protein tyrosine nitration. Our data reveal the existence of a relationship between ROS and 11 12 RNS metabolism under zinc stress and the contribution of nitro-oxidative stress to zinc sensitivity. The results also suggest that sensitivity to zinc is determined rather by the level of 13 oxidative than by the nitrosative processes in Brassica species. 14

15

16

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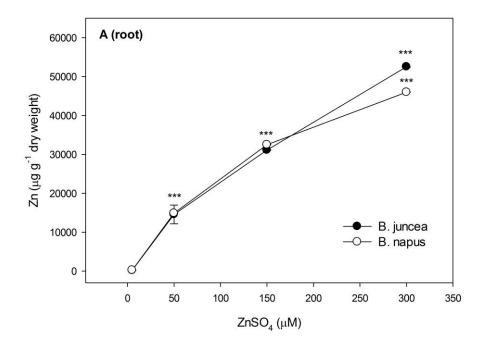
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1 Figures and figure legends

- 2 Fig 1 Zinc concentration (μg g⁻¹ DW) in roots (A) and shoots (B) of control and zinc-exposed
- 3 B. juncea and B. napus plants. Significant differences with respect to control plants (5 \square M
- 4 ZnSO₄) and according to Student's t-test (n = 10, *** $P \le 0.001$) are indicated.



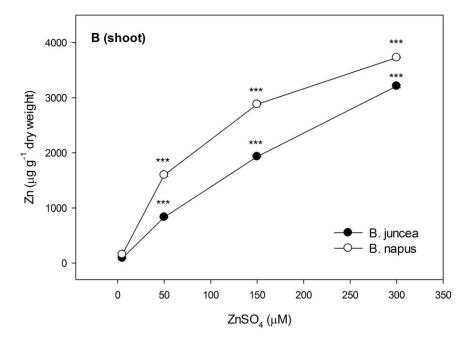


Fig 2 (A) Tissue specificity of zinc localization in control and zinc-treated *Brassica juncea* and *Brassica napus* root tips. Zinc localization was visualized with the help of Zinquin fluorophore as described in *Materials and methods*. Bar=200 μ m (B) Pixel intensities of Zinquin fluorescence measured in the PR meristem of control and Zn-treated *Brassica* species. Different letters indicate significant differences according to Duncan-test (n = 10, P \leq 0.05).

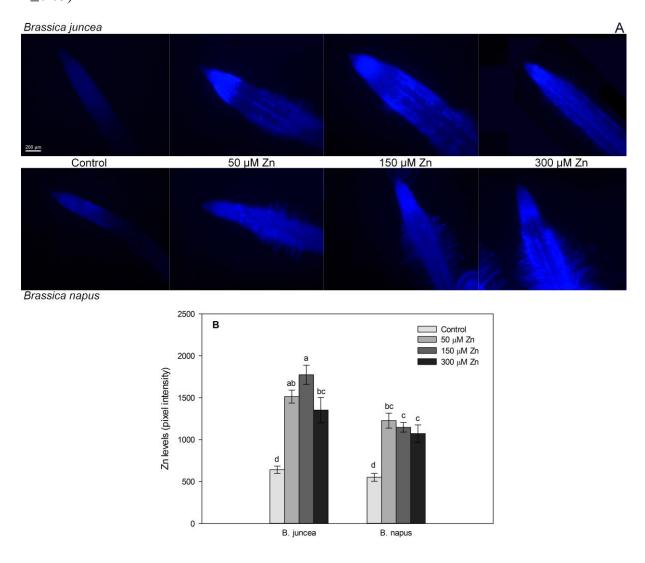
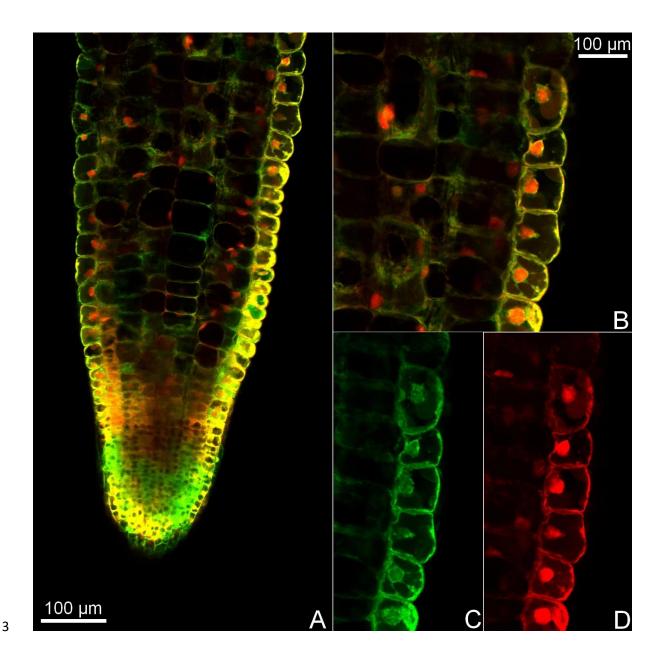


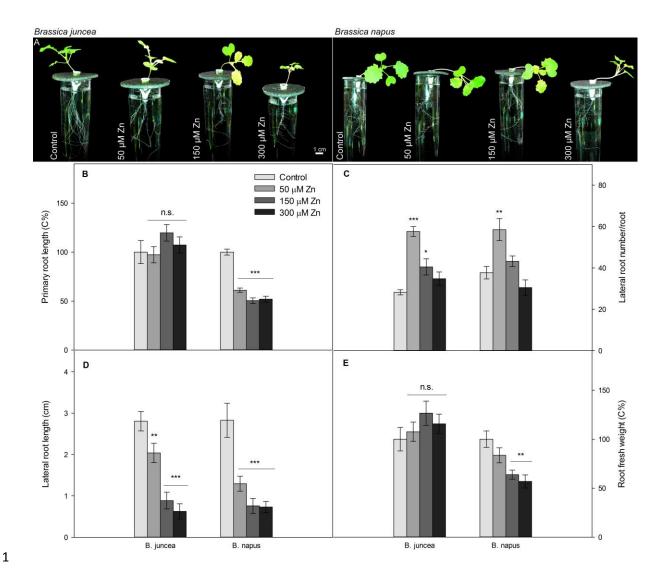
Fig 3 Confocal microscopic images of the root tip of 300 μM Zn-treated *B. juncea*. The root samples were co-stained with Zinpyr-1 and PI. (A) Lower magnification merged image of the whole root tip (A) and optically zoomed image showing root epidermal cells (B). Green

- 1 (ZinPyr) fluorescence corresponds to zinc (C) and red (PI) fluorescence shows cell walls and
- 2 nuclei (D). Bars=100 μm.



4 Fig 4 The effect of excess Zn on the growth of B. juncea and B. napus plants. (A)

- 5 Representative photographs of 16-days-old untreated and zinc-exposed *B. juncea* and *B.*
- 6 napus plants. Bar=1 cm. Primary root length in control % (B), number (C) and length (D) of
- 7 lateral roots and root fresh weight in control% (E) of *Brassica* plants treated with 0, 50, 150 or
- 8 300 μM Zn. The lack of significance (n.s.) or significant differences according to Student's t-
- 9 test $(n = 20, **P \le 0.01, ***P \le 0.001)$ are indicated.



2 Fig 5 Intensity of anilin blue-dependent fluorescence (in control %) reflecting callose levels in

- 3 the root tips of control and zinc-treated Brassica plants. Significant differences according to
- 4 Student's t-test (n = 10, *P \le 0.05, **P \le 0.01, ***P \le 0.001) are indicated.

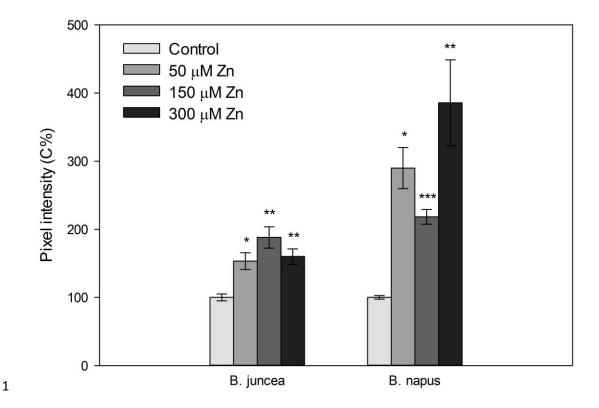


Fig 6 Viability (fluorescein pixel intensities in control %) of the root meristem cells of control and zinc-exposed *Brassica* plants. The lack of significance (n.s.) or significant differences according to Student's t-test (n = 10, *P≤0.05, **P≤0.01, ***P≤0.001) are indicated.

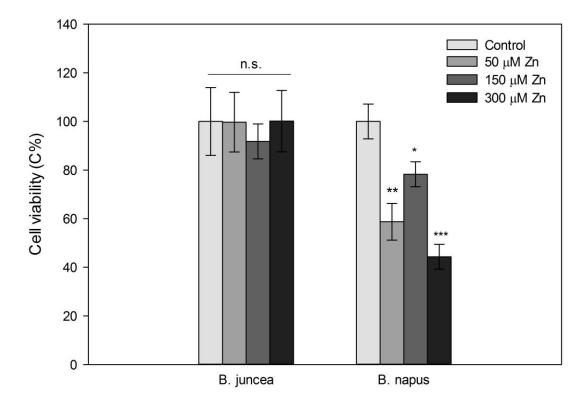
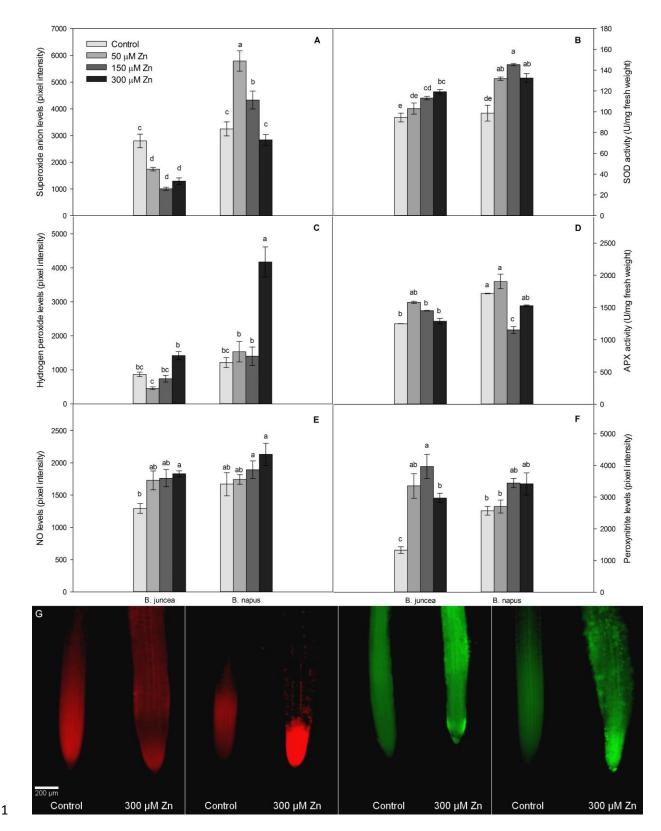


Fig 7 Effect of Zn excess on the metabolism of ROS and RNS. The levels of superoxide 1 radicals (pixel intensity of DHE fluorescence, A), total SOD activity (B), H₂O₂ (pixel 2 intensity of resorufin fluorescence, C), APX activity (D), NO (pixel intensity of DAF-FM 3 fluorescence, E) and ONOO (pixel intensity of APF fluorescence, F) in control and zinc-4 5 treated Brassica roots (in meristematic zone) are shown. (G) Representative microscopic images of Brassica root tips stained with different fluorophores (from left: DHE for 6 superoxide in B. napus, AmplifluTM for H₂O₂ in B. napus, DAF-FM for NO in B. napus, APF 7 8 for ONOO- in B. napus). The fluorescent staining procedures were carried out in the root tips, while enzymatic activities were measured in the whole root system as described in Materials 9 and methods. Different letters indicate significant differences according to Duncan-test (n = 10 10, P≤0.05). 11



2 Fig 8 Analysis of SOD isoforms. SOD isoforms were separated by native-PAGE and stained

3 by a photochemical method using 30 μg of protein per lane.

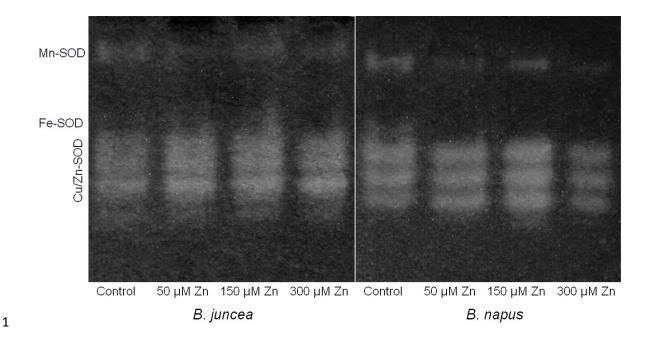
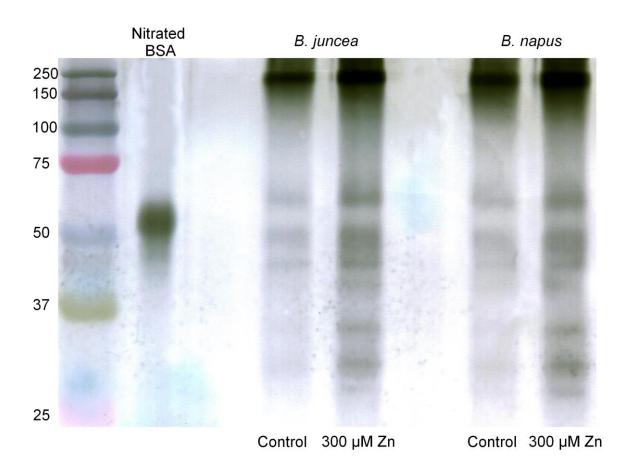


Fig 9 Representative immunoblots showing protein tyrosine nitration in roots of *B. juncea* and *B. napus* plants under control circumstances and subjected to 300 μM Zn. Root samples were separated by SDS-PAGE and analysed on Western blotting with anti-nitrotyrosine antibody (1:2000). Commercial nitrated BSA (NO₂-BSA) was used as a positive control.



- 1 Fig 10 Microscopic images of Brassica root tips stained with the Schiff's reagent. The pink
- 2 colorization indicates lipid peroxidation. Bar=200 μm.



- $\textbf{1} \quad \textbf{Table 1} \ \text{Copper (Cu), manganese (Mn) and iron (Fe) concentrations ($\mu g \cdot g^{\text{-1}}$ DW) in roots of}$
- 2 control and zinc-treated *Brassica* plants. Significant differences according to Student's t-test
- $(n = 10, ***P \le 0.001)$ are indicated.

Brassica juncea								
Control	50 μM Zn		150 μM Zn		$300 \mu M Zn$			
24.50 ± 0.38	173.00 ± 0.52		198.90 ± 0.49		174.80 ± 0.73			
250.00 ± 1.16	24.97 ± 0.05	* * *	25.53 ± 0.07	* *	26.02 ± 0.13	* * *		
519.10 ± 4.75	864.50 ± 3.65		869.90 ± 2.71	*	927.90 ± 2.22	*		
Brassica napus								
Control	50 μM Zn		150 μM Zn		$300 \mu M Zn$			
27.68 ± 0.08	150.60 ± 0.50		178.60 ± 0.85		163.70 ± 0.68			
145.10 ± 0.53	26.40 ± 0.09	* * *	22.02 ± 0.10		26.44 ± 0.17	* * *		
1051.0 ± 4.5	1496.0 ± 6.7		1540.0 ± 11.3		1239.0 ± 7.8	*		
	24.50 ± 0.38 250.00 ± 1.16 519.10 ± 4.75 $Control$ 27.68 ± 0.08 145.10 ± 0.53	Control50 μM Zn 24.50 ± 0.38 173.00 ± 0.52 250.00 ± 1.16 24.97 ± 0.05 519.10 ± 4.75 864.50 ± 3.65 Control $50 \mu M Zn$ 27.68 ± 0.08 150.60 ± 0.50 145.10 ± 0.53 26.40 ± 0.09	Control 50 μM Zn 24.50 ± 0.38 173.00 ± 0.52 250.00 ± 1.16 24.97 ± 0.05 * 519.10 ± 4.75 864.50 ± 3.65 Brassica Control $50 \mu M Zn$ 27.68 ± 0.08 150.60 ± 0.50 145.10 ± 0.53 26.40 ± 0.09 * ** *	Control 50 μM Zn 150 μM Zn 24.50 ± 0.38 173.00 ± 0.52 198.90 ± 0.49 250.00 ± 1.16 24.97 ± 0.05 * 25.53 ± 0.07 519.10 ± 4.75 864.50 ± 3.65 869.90 ± 2.71 Brassica napus Control $50 \mu M Zn$ $150 \mu M Zn$ 27.68 ± 0.08 150.60 ± 0.50 178.60 ± 0.85 145.10 ± 0.53 26.40 ± 0.09 * 22.02 ± 0.10	Control 50 μM Zn 150 μM Zn 24.50 ± 0.38 173.00 ± 0.52 198.90 ± 0.49 250.00 ± 1.16 24.97 ± 0.05 * 25.53 ± 0.07 * 519.10 ± 4.75 864.50 ± 3.65 869.90 ± 2.71 Brassica napus Control $50 \mu M Zn$ $150 \mu M Zn$ 27.68 ± 0.08 150.60 ± 0.50 178.60 ± 0.85 145.10 ± 0.53 26.40 ± 0.09 * 22.02 ± 0.10	Control 50 μM Zn 150 μM Zn 300 μM Zn 24.50 ± 0.38 173.00 ± 0.52 198.90 ± 0.49 174.80 ± 0.73 250.00 ± 1.16 24.97 ± 0.05 * 25.53 ± 0.07 * 26.02 ± 0.13 519.10 ± 4.75 864.50 ± 3.65 869.90 ± 2.71 927.90 ± 2.22 Brassica napus Control 50 μM Zn 150 μM Zn 300 μM Zn 27.68 ± 0.08 150.60 ± 0.50 178.60 ± 0.85 163.70 ± 0.68 145.10 ± 0.53 26.40 ± 0.09 * 22.02 ± 0.10 26.44 ± 0.17		