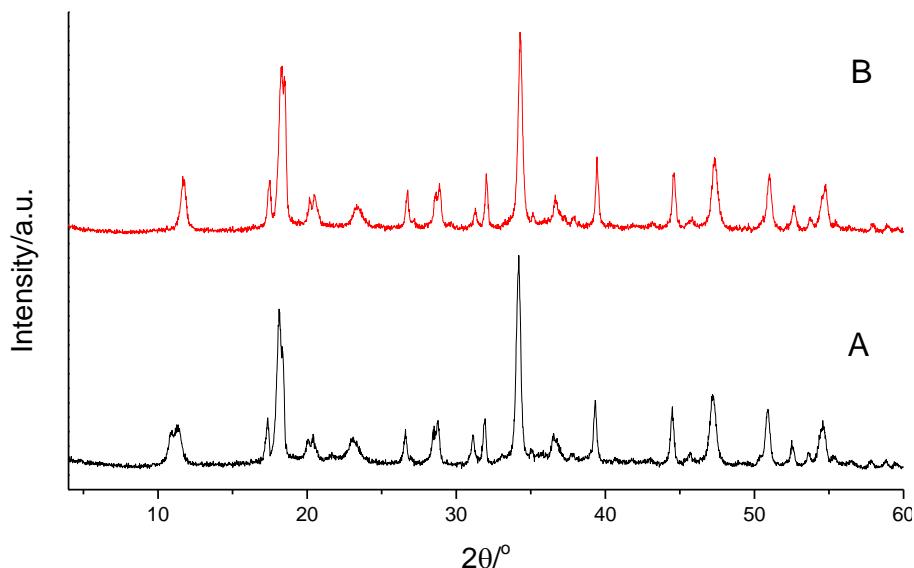


## Supporting information

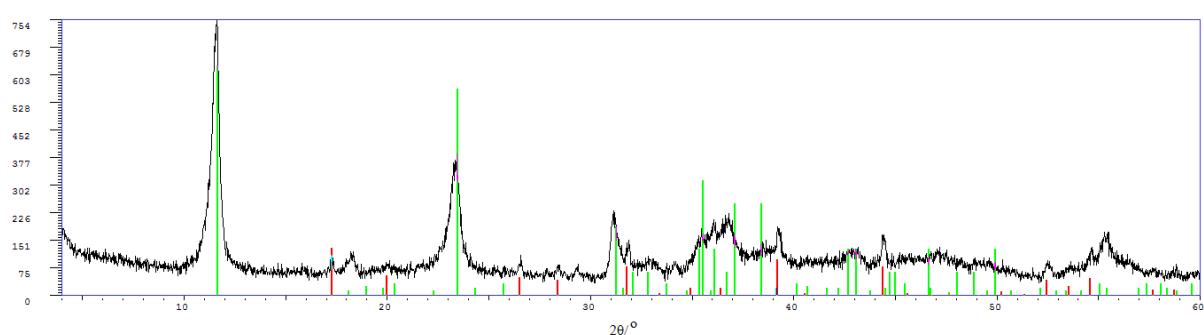
to

Optimisation of the synthesis parameters of mechanochemically prepared CaAl-layered double hydroxide

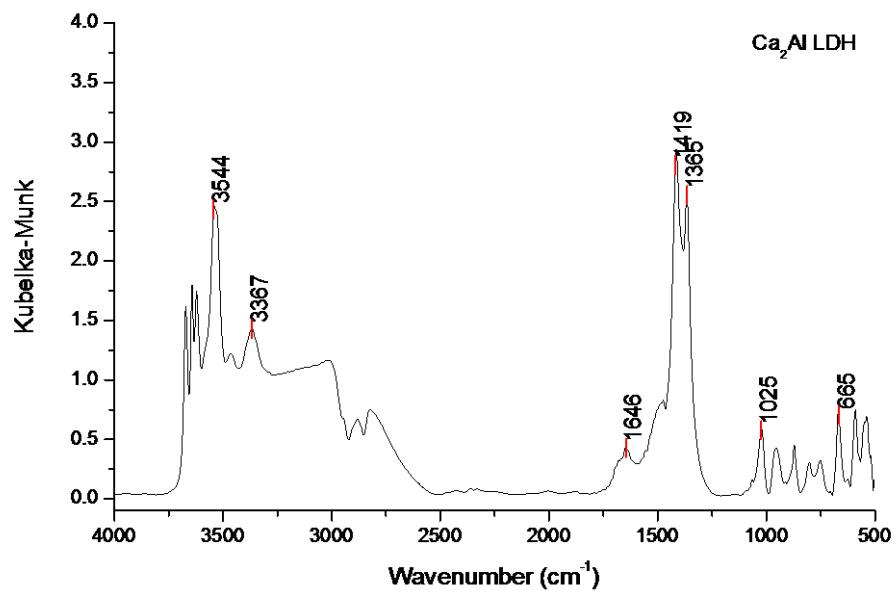
Zsolt Ferencz, Ákos Kukovecz, Zoltán Kónya, Pál Sipos, István Pálinkó



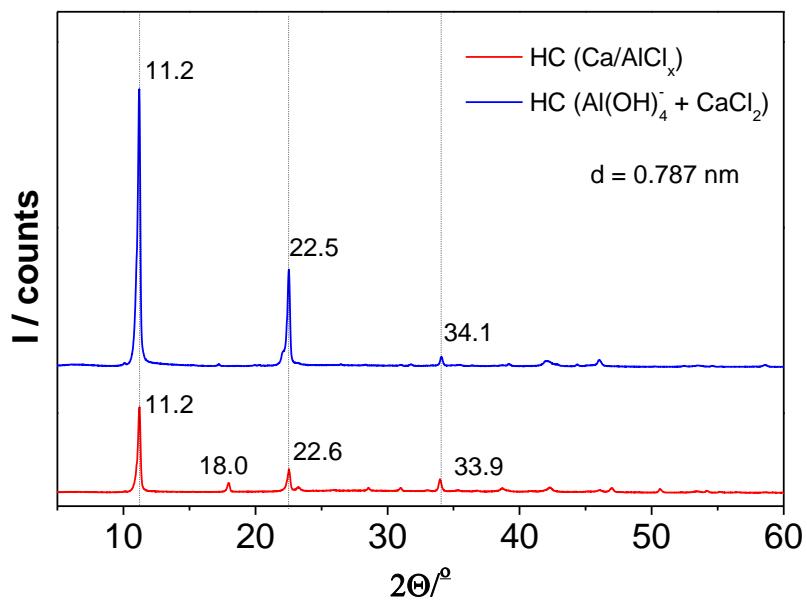
**Fig. S1.** The powder X-ray diffractograms of the  $\text{Ca}_2\text{Al}_{-60}$  ( $\text{B}/\text{S} = 60$ ) sample (A) before and (B) after washing.



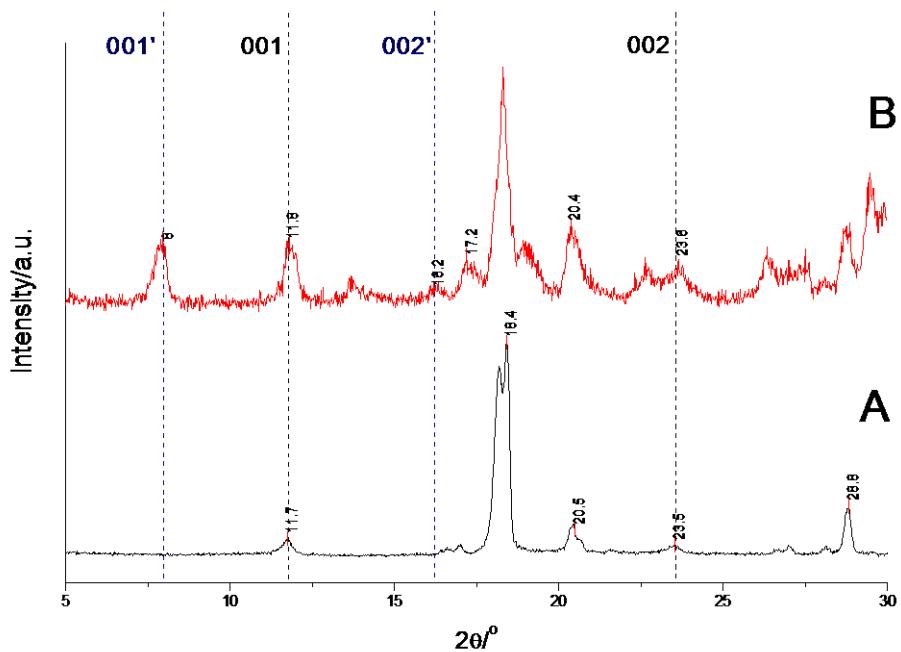
**Fig. S2.** The phase composition of the  $\text{Ca}_2\text{Al}$  LDH sample prepared with the optimised parameters:  $\text{B}/\text{S} = 140$ ,  $n(\text{H}_2\text{O}):n(\text{Al}) = 1,75$  and  $v = 11.6$  Hz. Black curve: the sample, green line: the diffraction pattern of the JCPDS #41-0219, red line: the diffraction pattern of the JCPDS #74-2281 (calcium-aluminium hydroxide).



**Fig. S3.** FT-IR spectrum of the  $\text{Ca}_2\text{Al}$ -LDH prepared under the optimised conditions (see the vibration  $1365 \text{ cm}^{-1}$ , typical of interlayer carbonate ion).



**Fig. S4.** The powder XRD diffraction patterns of two hydrocalumite samples (red:  $\text{NaOH} + \text{CaCl}_2/\text{AlCl}_3$ ; blue:  $\text{Al(OH)}_4^-/\text{NaOH} + \text{CaCl}_2$ ) prepared by the co-precipitation technique (Tóth et al., 2014).



**Fig. S5.** Powder X-ray diffractograms of (A) the pristine  $\text{Ca}_2\text{Al}_1$ -LDH and (B), its tyrosinate intercalated derivative (B).

### Reference

Tóth, V., Sipiczki, M., Pallagi, A., Kukovecz, A., Kónya, Z., Sipos, P., Pálinkó, I., 2014, Synthesis and properties of CaAl-layered double hydroxides of hydrocalumite-type, 68, 633–637.