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Letter

Pedrolane, a Polycyclic Diterpene Scaffold Containing a Bicyclo[2.2.1]heptane System, from *Euphorbia pedroi*

Ricardo J. Ferreira, Gabriella Spengler, Andreas Orthaber, Daniel J. V. A. dos Santos, and Maria-José U. Ferreira*



ABSTRACT: Pedrolide (1), a diterpenoid with an unprecedented carbon skeleton, pedrolane, containing a bicycle[2.2.1]heptane system, is reported. This structural feature is hypothesized to involve an intramolecular cyclization, via Michael addition, and a ring contraction, via 1,2-alkyl shift or a Pinacol rearrangement of rings A and B, from a tigliane-type 5/7/6/3-tetracyclic ring precursor. The structure of 1 was established using spectroscopic techniques, single-crystal X-ray diffraction, and *ab initio* calculations. Pedrolide reversed multidrug resistance mediated by P-glycoprotein.

The genus *Euphorbia* (Euphorbiaceae) is a key source of isoprenoids with unique structural diversity and remarkable biological activities. Most particularly, polyoxygenated macrocyclic jatrophane and lathyrane-type diterpenes and the biogenetically related polycyclic diterpenes, bearing tigliane, ingenane, and daphnane scaffolds, generally found as aliphatic and aromatic acyl derivatives, are considered important groups of compounds due to their multidrug-resistance reversal (MDR), cytotoxic, antitumor, antiviral, and anti-inflammatory activities.¹

Among these diterpenes, exclusively found in Euphorbiaceae and Thymelaeaceae families, the polycylic diterpene tigilanol tiglate, for treating solid tumors,² and the daphnane-type diterpene resiniferatoxin, a strong nonopioid analgesic able to provide pain relief in patients with advanced cancer,³ are currently in clinical trials and represent two examples of the therapeutic potential of diterpenes from *Euphorbia* species. Tiglianes are characterized by a 5/7/6/3-tetracyclic ring system (A/B/C/D), with the fusion of rings A–B usually *trans*, although the *cis* fusion of rings A–B is increasingly being reported.^{1,4,5} Unvaryingly, the fusion of rings B–C is always *trans* and of C–D rings *cis*. A variable oxidation pattern, regarding the position of the double bonds and acylation pattern of oxygen functions, render them an unusual structural chemical diversity. Tiglianes are also characterized by the presence of an α,β -unsaturated carbonyl system at ring A, which is reported to be responsible for most of the activities described for this class of compounds, including proinflammatory, cocarcinogenic, tumorigenic, anticancer and antiviral activities.^{1,6}

In our search for effective MDR reversers,^{7–12} the phytochemical study of *Euphorbia pedroi* Molero and Rovira, an endemic plant of the western sea cliffs in Portugal, was performed.⁷ Herein a novel noncytotoxic tigliane-type diterpenoid was isolated from the methanol extract, with a previously unreported skeleton, which was evaluated for its ability to inhibit P-glycoprotein (P-gp/ABCB1).

Compound 1, named pedrolide, was isolated as white needles with $[\alpha]_D^{20}$ +75.4° (*c* 0.1, CHCl₃). While the IR spectrum provided evidence for the carbonyl function (1750 and 1701 cm⁻¹), the low-resolution molecular ESI-MS spectrum revealed the presence of a protonated molecular

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ion at m/z 521 [M + H]⁺. From the ESI-HRMS spectrum, the identification of a potassium adduct ion at m/z 559.2096 [M + K]⁺ allowed us to infer a molecular formula of C₃₁H₃₆O₇ (Calcd for C₃₁H₃₆O₇K, 559.2093), along with a degree of unsaturation of 14.

The ¹H NMR spectrum (Table 1) showed proton resonances corresponding to seven methyl groups, three

Table 1. ¹H and ¹³C NMR Spectra for Pedrolide (1) (CDCl₃, 300 and 75 MHz, Respectively)

1 55.6 $2.63 (m)$ 2 45.0 $2.39 (qd, 7.2, 4.7)$ 3 212.2 4 63.1 $2.61 (br d, 2.1)$ 5 176.8 6 45.6 7 46.7 $2.44 (br s)$ 8 51.0 $1.75 (br d, 5.5)$ 9 88.5 10 50.1 $3.03 (dd, 4.5, 2.1)$ 11 43.4 $1.80 (dq, 8.2, 7.0)$ 12 80.1 $5.95 (d, 8.2)$ 13 66.4 14 28.2 $0.96 (d, 5.5)$ 15 30.3 16 23.2 $1.12 (s)$ 17 18.0 $1.45 (s)$ 18 13.8 $1.18 (d, 7.0)$ 19 12.4 $1.25 (d, 7.2)$ 20 17.4 $1.25 (s)$ 1' 166.3 2' 130.9 3' 129.8 $8.02 (dd, 7.2, 1.4)$ 4' 128.5 $7.44 (dd, 7.2, 1.4)$ 4' 128.5 $7.44 (dd, 7.2, 1.4)$ 5' 132.8 $7.55 (dd, 7.2, 1.4)$ 1'' 177.3 $2''$ 2'' 33.8 $2.54 (h, 7.0)$ 3'' 19.1 $1.12 (d, 7.0)$	position	$\delta_{ m C}$	$\delta_{ m H}$ (mult, J in Hz)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	1	55.6	2.63 (m)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	2	45.0	2.39 (qd, 7.2, 4.7)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	3	212.2	
5176.8645.6746.72.44 (br s)851.01.75 (br d, 5.5)988.51050.13.03 (dd, 4.5, 2.1)1143.41.80 (dq, 8.2, 7.0)1280.15.95 (d, 8.2)1366.41428.20.96 (d, 5.5)1530.31623.21.12 (s)1718.01.45 (s)1813.81.18 (d, 7.0)1912.41.25 (d, 7.2)2017.41.25 (s)1'166.32'130.93'129.88.02 (dd, 7.2, 1.4)4'128.57.44 (dd, 7.2, 1.4)5'132.87.55 (dd, 7.2, 1.4)1"177.32"3.82.54 (h, 7.0)3"19.11.12 (d, 7.0)	4	63.1	2.61 (br d, 2.1)
6 45.6 7 46.7 2.44 (br s)8 51.0 1.75 (br d, 5.5)9 88.5 10 50.1 3.03 (dd, 4.5 , 2.1)11 43.4 1.80 (dq, 8.2 , 7.0)12 80.1 5.95 (d, 8.2)13 66.4 14 28.2 0.96 (d, 5.5)15 30.3 16 23.2 1.12 (s)17 18.0 1.45 (s)18 13.8 1.18 (d, 7.0)19 12.4 1.25 (d, 7.2)20 17.4 1.25 (s)1' 166.3 2' 130.9 3' 129.8 8.02 (dd, 7.2 , 1.4)4' 128.5 7.44 (dd, 7.2 , 1.4)5' 132.8 7.55 (dd, 7.2 , 1.4)1'' 177.3 $2''$ 2'' 33.8 2.54 (h, 7.0)3'' 19.1 1.12 (d, 7.0)	5	176.8	
746.72.44 (br s)851.01.75 (br d, 5.5)988.51050.13.03 (dd, 4.5, 2.1)1143.41.80 (dq, 8.2, 7.0)1280.15.95 (d, 8.2)1366.41428.20.96 (d, 5.5)1530.31623.21.12 (s)1718.01.45 (s)1813.81.18 (d, 7.0)1912.41.25 (d, 7.2)2017.41.25 (s)1'166.32'130.93'129.88.02 (dd, 7.2, 1.4)4'128.57.44 (dd, 7.2, 1.4)5'132.87.55 (dd, 7.2, 1.4)1''177.32''2''33.82.54 (h, 7.0)3''19.11.12 (d, 7.0)	6	45.6	
8 51.0 1.75 (br d, 5.5)988.510 50.1 3.03 (dd, 4.5 , 2.1)11 43.4 1.80 (dq, 8.2 , 7.0)12 80.1 5.95 (d, 8.2)13 66.4 14 28.2 0.96 (d, 5.5)15 30.3 16 23.2 1.12 (s)17 18.0 1.45 (s)18 13.8 1.18 (d, 7.0)19 12.4 1.25 (d, 7.2)20 17.4 1.25 (s)1' 166.3 2' 130.9 3' 129.8 8.02 (dd, 7.2 , 1.4)4' 128.5 7.44 (dd, 7.2 , 1.4)5' 132.8 7.55 (dd, 7.2 , 1.4)1'' 177.3 $2''$ 2'' 33.8 2.54 (h, 7.0)3'' 19.1 1.12 (d, 7.0)	7	46.7	2.44 (br s)
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10 50.1 $3.03 (dd, 4.5, 2.1)$ 11 43.4 $1.80 (dq, 8.2, 7.0)$ 12 80.1 $5.95 (d, 8.2)$ 13 66.4 14 28.2 $0.96 (d, 5.5)$ 15 30.3 16 23.2 $1.12 (s)$ 17 18.0 $1.45 (s)$ 18 13.8 $1.18 (d, 7.0)$ 19 12.4 $1.25 (d, 7.2)$ 20 17.4 $1.25 (s)$ 1' 166.3 2' 130.9 3' 129.8 $8.02 (dd, 7.2, 1.4)$ 4' 128.5 $7.44 (dd, 7.2, 1.4)$ 5' 132.8 $7.55 (dd, 7.2, 1.4)$ 1'' 177.3 2'' 33.8 $2.54 (h, 7.0)$ 3'' 19.1 $1.12 (d, 7.0)$	9	88.5	
1143.41.80 (dq, 8.2, 7.0)1280.1 5.95 (d, 8.2)1366.41428.2 0.96 (d, 5.5)1530.31623.2 1.12 (s)1718.0 1.45 (s)1813.8 1.18 (d, 7.0)1912.4 1.25 (d, 7.2)2017.4 1.25 (s)1'166.32'130.93'129.8 8.02 (dd, 7.2, 1.4)4'128.5 7.44 (dd, 7.2, 1.4)5'132.8 7.55 (dd, 7.2, 1.4)1''177.32''33.8 2.54 (h, 7.0)3''19.1 1.12 (d, 7.0)	10	50.1	3.03 (dd, 4.5, 2.1)
1280.1 $5.95 (d, 8.2)$ 13 66.4 14 28.2 $0.96 (d, 5.5)$ 15 30.3 16 23.2 $1.12 (s)$ 17 18.0 $1.45 (s)$ 18 13.8 $1.18 (d, 7.0)$ 19 12.4 $1.25 (d, 7.2)$ 20 17.4 $1.25 (s)$ 1' 166.3 2' 130.9 3' 129.8 $8.02 (dd, 7.2, 1.4)$ 4' 128.5 $7.44 (dd, 7.2, 1.4)$ 5' 132.8 $7.55 (dd, 7.2, 1.4)$ 1'' 177.3 2'' 33.8 $2.54 (h, 7.0)$ 3'' 19.1 $1.12 (d, 7.0)$	11	43.4	1.80 (dq, 8.2, 7.0)
13 66.4 14 28.2 $0.96 (d, 5.5)$ 15 30.3 16 23.2 $1.12 (s)$ 17 18.0 $1.45 (s)$ 18 13.8 $1.18 (d, 7.0)$ 19 12.4 $1.25 (d, 7.2)$ 20 17.4 $1.25 (s)$ 1' 166.3 2' 130.9 3' 129.8 $8.02 (dd, 7.2, 1.4)$ 4' 128.5 $7.44 (dd, 7.2, 1.4)$ 5' 132.8 $7.55 (dd, 7.2, 1.4)$ 1'' 177.3 2'' 33.8 $2.54 (h, 7.0)$ 3'' 19.1 $1.12 (d, 7.0)$	12	80.1	5.95 (d, 8.2)
14 28.2 0.96 (d, 5.5) 15 30.3 16 23.2 1.12 (s) 17 18.0 1.45 (s) 18 13.8 1.18 (d, 7.0) 19 12.4 1.25 (d, 7.2) 20 17.4 1.25 (s) $1'$ 166.3 $2'$ 130.9 $3'$ 129.8 8.02 (dd, 7.2, 1.4) $4'$ 128.5 7.44 (dd, 7.2, 1.4) $5'$ 132.8 7.55 (dd, 7.2, 1.4) $1''$ 177.3 $2''$ 33.8 2.54 (h, 7.0) $3''$ 19.1 1.12 (d, 7.0)	13	66.4	
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1623.21.12 (s)1718.01.45 (s)1813.81.18 (d, 7.0)1912.41.25 (d, 7.2)2017.41.25 (s)1'166.32'130.93'129.88.02 (dd, 7.2, 1.4)4'128.57.44 (dd, 7.2, 1.4)5'132.87.55 (dd, 7.2, 1.4)1"177.32"33.82.54 (h, 7.0)3"19.11.12 (d, 7.0)	15	30.3	
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1813.81.18 (d, 7.0)1912.41.25 (d, 7.2)2017.41.25 (s)1'166.32'130.93'129.88.02 (dd, 7.2, 1.4)4'128.57.44 (dd, 7.2, 1.4)5'132.87.55 (dd, 7.2, 1.4)1"177.32"33.82.54 (h, 7.0)3"19.11.12 (d, 7.0)	17	18.0	1.45 (s)
1912.41.25 (d, 7.2)2017.41.25 (s)1'166.32'130.93'129.88.02 (dd, 7.2, 1.4)4'128.57.44 (dd, 7.2, 1.4)5'132.87.55 (dd, 7.2, 1.4)1''177.32''33.82.54 (h, 7.0)3''19.11.12 (d, 7.0)	18	13.8	1.18 (d, 7.0)
20 17.4 1.25 (s)1' 166.3 2' 130.9 3' 129.8 8.02 (dd, 7.2 , 1.4)4' 128.5 7.44 (dd, 7.2 , 1.4)5' 132.8 7.55 (dd, 7.2 , 1.4)1'' 177.3 2'' 33.8 2.54 (h, 7.0)3'' 19.1 1.12 (d, 7.0)	19	12.4	1.25 (d, 7.2)
1'166.3 $2'$ 130.9 $3'$ 129.88.02 (dd, 7.2, 1.4) $4'$ 128.57.44 (dd, 7.2, 1.4) $5'$ 132.87.55 (dd, 7.2, 1.4) $1''$ 177.3 $2''$ 33.82.54 (h, 7.0) $3''$ 19.11.12 (d, 7.0)	20	17.4	1.25 (s)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	1'	166.3	
3'129.88.02 (dd, 7.2, 1.4) $4'$ 128.57.44 (dd, 7.2, 1.4) $5'$ 132.87.55 (dd, 7.2, 1.4) $1''$ 177.3 $2''$ 33.82.54 (h, 7.0) $3''$ 19.11.12 (d, 7.0)	2'	130.9	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	3'	129.8	8.02 (dd, 7.2, 1.4)
5' 132.8 7.55 (dd, 7.2, 1.4) 1" 177.3 2" 33.8 2.54 (h, 7.0) 3" 19.1 1.12 (d, 7.0)	4′	128.5	7.44 (dd, 7.2, 1.4)
1" 177.3 2" 33.8 2.54 (h, 7.0) 3" 19.1 1.12 (d, 7.0)	5'	132.8	7.55 (dd, 7.2, 1.4)
2" 33.8 2.54 (h, 7.0) 3" 19.1 1.12 (d, 7.0)	1″	177.3	
3″ 19.1 1.12 (d, 7.0)	2″	33.8	2.54 (h, 7.0)
	3″	19.1	1.12 (d, 7.0)
4" 18.9 1.10 (d, 7.0)	4″	18.9	1.10 (d, 7.0)

tertiary, as singlets, at $\delta_{\rm H}$ 1.12, 1.25, and 1.45 and four doublets of secondary methyl groups at $\delta_{\rm H}$ 1.10 (J = 7.0 Hz), 1.12 (J = 7.0 Hz), 1.18 (J = 7.0 Hz), and 1.25 (J = 7.2 Hz) and a downfield signal of an oxymethine at $\delta_{\rm H}$ 5.95 (1H, d, J = 8.2 Hz). Additionally, a multiplet at $\delta_{\rm H}$ 2.54 (h, J = 7.0 Hz) in the aliphatic region indicated the presence of an isobutyryl group. Moreover, resonances corresponding to five aromatic protons ($\delta_{\rm H}$ 7.44, 7.55, and 8.02) were also observed, suggesting the presence of a benzoyl moiety. Both acyl groups were substantiated by the ¹³C NMR data (isobutyryl, carbonyl at $\delta_{\rm C}$ 177.3 and two methyls and one methine in the aliphatic region; benzoyl, carbonyl at $\delta_{\rm C}$ 166.3 and six aromatic carbons).

In addition to the signals of the isobutyryl and benzoyl esters, the remaining signals in the ¹³C NMR spectrum, along with the DEPT experiment, revealed the presence of twenty carbons, namely, five methyl groups, nine methines (one oxygenated, at $\delta_{\rm C}$ 80.1), and six quaternary carbons, including two carbonyl (ketone at $\delta_{\rm C}$ 212.2 and one acyl function at $\delta_{\rm C}$ 176.8) and two oxygenated carbons ($\delta_{\rm C}$ 66.4 and 88.5). The above structural features pointed to a diterpene scaffold for 1.

Detailed analysis of the two-dimensional NMR experiments $({}^{1}H{-}^{1}H$ COSY, HMQC, HMBC, and NOESY) allowed us to elucidate the relevant structural details of 1, further assign proton and carbon signals, and locate the functional groups (Figure 1).



Figure 1. Structure and key COSY, HMBC, and NOESY correlations for pedrolide (1). The new structural features of compound (1) are depicted in blue. Distances between ¹H atom pairs, in angstroms, are depicted above the corresponding NOE correlation.

The ${}^{1}H-{}^{1}H$ COSY and HMQC spectra indicated three spin systems (A-C, Figure 1) connected by long-range heteronuclear correlations, found in the HMBC spectrum. A HMBC $^{2}J_{C-H}$ correlation between H-10 and C-9, and a $^{3}J_{C-H}$ correlation between Me-18 and C-9 connected the spin system A to B. Following, ${}^{3}J_{C-H}$ correlations between H-7/Me-20, H-1/C-6, and H-10/C-6 located C-6 as bridging C-4 and C-7. In turn, correlations between H-12/C-13, H-14/C-13/C15, Me-16/C-13/C-14/C-15, and Me-17/C-13/C-15 connected spin system B to the cyclopropane ring. Concerning the position of the functional groups, the correlations found between Me-19 and C-2/C-3 and H-2/C-3 located the ketone group at C-3 ($\delta_{\rm C}$ 212.2). The lactone ($\delta_{\rm C}$ 176.8), between C-6 ($\delta_{\rm C}$ 45.6) and C-9 ($\delta_{\rm C}$ 88.5), was substantiated by the $^2J_{\rm C-H}$ correlations of Me-20 with C-6 ($\delta_{\rm C}$ 45.6) and ${}^3\!J_{\rm C-H}$ correlations of Me-20 with C-5 ($\delta_{\rm C}$ 176.8) and C-7 ($\delta_{\rm C}$ 46.7). The benzoate ester was placed at C-12 based on a ${}^{3}J_{\rm C-H}$ correlation between H-12 $(\delta_{\rm H} 5.95)$ and the carbonyl carbon at $\delta_{\rm C}$ 166.3. The absence of

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HMBC correlations involving the carbonyl at δ_C 179.6 of the isobutyrate ester function located it at C-13 (δ_C 66.4).^{13,14}

The above data provided evidence for the presence of some structural features, namely, ring C and the cyclopropane ring D, of the tigliane scaffold in the structure of 1. However, other typical elements as the enone at ring A and a double bond at ring B (usually at C-5 or C-6) were absent. In fact, a detailed analysis of the $^{1}H-^{1}H$ COSY data and HMBC correlations observed for H-1, H-4, H-7, H-8, and H-10 indicated the occurrence of a rearrangement involving this particular region of the molecule, which was assigned as a bicyclo[2.2.1]heptane feature.

To substantiate the type of rearrangement that occurred in 1, single crystal diffraction data (CCDC 1990457) of 1 was obtained (Figure 2). It allowed us to confirm that pedrolide



Figure 2. Single-crystal X-ray structure of 1.

(1) was, in fact, a new compound with an unprecedented carbon skeleton, herein named pedrolane. However, and due to the weakly diffracting nature of the crystals, no information could be retrieved regarding the absolute configuration of 1.

The relative stereochemistry of 1 was thus assigned through both a NOESY experiment and an ab initio energy minimization. After importing the heavy atoms' spatial coordinates, obtained from the crystallographic structure into MOE, we performed two rounds of energy minimization to confirm the orientation of the hydrogen atoms (Supporting Information, Figure S20). Afterward, we cross-referenced the obtained spatial distances (Figure 1) with the 2D NOESY data. Accordingly, NOE cross peaks between H-14/Me-16, H-14/ H-7, H-7/Me-20, H-7/Me-19, and Me-20/H-4 showed that these protons are on the same side of the molecule. In turn, NOE correlations between H-11/Me-17, Me-17/H-8, H-8/H-1 (very strong), H-1/H-2, H-1/H-10, and H-10/H-2, indicated that these protons are on the opposite side of the molecule. Moreover, the configuration at the stereocenters C-11 and C-12 were corroborated by NOE correlations between Me-18 and H-12.

Thus, from the 2D NOESY data we obtained a relative configuration that was further used as input to calculate the specific rotation via an *ab initio* approach¹⁵ (Supporting Information, Computational Methodology section). We calculated the specific rotation of both epimers 4R and 4S for comparison purposes. Herein, the calculated $[\alpha]_D$ values

were +84.0° and -80.9° for epimers 4S and 4R, respectively. Therefore, and due to the very good agreement of epimer 4S with the experimentally determined specific rotation (+75.4°), the absolute configuration of pedrolide (1) is proposed to be 1*S*,2*S*,4*S*,6*S*,7*R*,8*S*,9*R*,10*R*,11*R*, 12*R*,13*S*,14*R*.

A hypothetical biosynthetic pathway for the carbon scaffold of compound 1 is proposed (Scheme 1) starting from a





precursor with a tetracyclic diterpene scaffold of the tiglianetype (5/7/6/3 ring system; Scheme 1, A), bearing some characteristic features as an α , β -unsaturated carbonyl at C-3, hydroxyl groups at C-9, C-12, and C-13, and a double bond at C-6. Furthermore, the observed chemical shift for H-8 ($\delta_{\rm H}$ 1.75) additionally suggested that this precursor had their rings A–B in a *cis*-fused configuration.¹

Accordingly, a Michael addition reaction of the double bond at C-6, with the β -carbon of the α , β -unsaturated system, leads to intramolecular cyclization, by connecting C-1 with C-7 (Scheme 1, (i)). Next, two simultaneous steps may occur involving a ring contraction, originating the new carbon– carbon bond C-4/C-6, via a 1,2-alkyl shift through C-4/C-5 Cbond cleavage, and water attack at C-5 (Scheme 1, (ii)). The oxidation of the resulting hydroxymethyl at C-5 into a carboxylic acid (Scheme 1, (iii)) would favor a nucleophilic attack by the hydroxyl group at C-9, therefore originating the lactone ring (Scheme 1, (iv)).

Alternatively, alkene formation at C-5, epoxidation, epoxide hydrolysis, leading to a vicinal diol, followed by Pinacol rearrangement would generate an aldehyde at C-5, which could be further oxidized to carboxylic acid (Supporting Information, Scheme S1).

Further acylations at C-12 and C-13 (Scheme 1, (v)) would form pedrolide (1), bearing the new pedrolane-type scaffold.

As it has been proven that orbital interactions have a critical role in the binding of nucleophiles to Michael acceptors,¹⁶ *ab initio* calculations were additionally performed. The obtained results support our hypothesis for the first step of proposed biogenetic pathway, only favored when rings A–B are in a *cis*-fused configuration (Supporting Information, Figures S21 and S22).

The cytotoxic activity of **1** was evaluated in resistant mouse *ABCB1*-transfected T-lymphoma (L5178Y-MDR) cells and corresponding sensitive subline (L5178Y-PAR), in resistant (Colo320) and sensitive human colon adenocarcinoma cells (Colo205), and in sensitive murine (NIH/3T3) and human (MRC-5) embryonic fibroblast cell lines by the thiazolyl blue bromide (MTT) assay.^{17,18} No relevant cytotoxic effects were found for compound **1** in any of the above cell lines, with IC₅₀ values ranging from 73.3 μ M to more than 100 μ M, (Supporting Information, Table S3).

Multidrug resistance (MDR) to anticancer drugs is the major impediment in the successful treatment of cancer. MDR is frequently associated with the overexpression of ABCtransporter proteins, responsible for removing anticancer drugs from cancer cells, being P-glycoprotein the most implicated. The development of P-gp inhibitors, able to impair the efflux activity of this ABC-transporter, is one of the most promising approaches to overcome MDR in cancer.¹⁹ The ability of compound 1 to inhibit P-gp efflux was assessed in resistant mouse lymphoma (L5178Y-MDR) and human colon adenocarcinoma (Colo320) cell lines through the rhodamine-123 accumulation assay, at the concentrations of 2 and 20 μ M and using verapamil as a positive control. Herein, an active compound was defined as having a fluorescence activity ratio (FAR) > 1.0, meaning that the compound was able to impair rhodamine-123 efflux by P-gp. As shown in Figure S13, pedrolide (1) was active in both cell lines (L5178Y-MDR cells, FAR 1.87 and 16.04 at 2 and 20 μ M, respectively; Colo320, FAR 1.64 and 5.36 at 2 and 20 μ M, respectively) and in both cases was found to have a greater activity than verapamil at 20 μ M (FAR = 9.66 and 4.10 for L5178Y-MDR and Colo320 cells, respectively). As indicated by the absence of significant changes in size (FSC) and granularity (SSC) of the cell population (Supporting Information, Figures S14-S19), toxicity effects could also be excluded in the flow cytometry assav.

Compound 1 was also evaluated in combination with the cytotoxic drug doxorubicin, one of the most used anticancer drugs in chemotherapy and a known P-gp substrate.¹⁹ The extent of interaction between doxorubicin and the compound was calculated by a combination index (CI) as suggested by Chou. The combination index can be estimated using the multiple drug effect equation derived from the median-effect principle of the law of mass-action, providing a quantitative assessment for synergistic (CI < 1), additive (CI = 1), and antagonist (CI > 1) effects. Furthermore, both the potency and shape of the dose-effect curve of each drug alone (and their combination) are also taken into consideration.^{20,21} Thus, any compound able to inhibit P-gp is expected to potentiate cytotoxic effects when coadministered with any given anticancer drug. From the results in Table S4 (Supporting Information) it can be concluded that compound 1 showed a strong synergism (CI = 0.208) with doxorubicin, enhancing its cytotoxicity in resistant L5178Y-MDR cells by more than 6fold (IC $_{50}$ = 0.396 μM , against IC $_{50}$ = 2.49 μM of doxorubicin alone).

ASSOCIATED CONTENT

3 Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.orglett.0c03647.

- Material and Methods, IR, MS, and 1D- and 2D-NMR, crystal data/structure refinement, biological activities, and computational details (PDF)
- Gaussian16 output files concerning: energy minimization of pedrolide, optical rotation calculations for epimers 4*R* and 4*S*, and *ab initio* calculations concerning the energetics of first step of the biogenetic pathway for *cis-* and *trans-*fused hypothetical precursor (ZIP)

Accession Codes

CCDC 1990457 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

AUTHOR INFORMATION

Corresponding Author

Maria-José U. Ferreira – Research Institute for Medicines (iMed.ULisboa), Faculty of Pharmacy, Universidade de Lisboa, 1649-003 Lisbon, Portugal; orcid.org/0000-0002-8742-1486; Phone: +351 217946475; Email: mjuferreira@ff.ulisboa.pt; Fax: +351 217946470

Authors

- Ricardo J. Ferreira Research Institute for Medicines (iMed.ULisboa), Faculty of Pharmacy, Universidade de Lisboa, 1649-003 Lisbon, Portugal; © orcid.org/0000-0003-2590-8229
- Gabriella Spengler Department of Medical Microbiology and Immunobiology, Faculty of Medicine, University of Szeged, H-6720 Szeged, Hungary
- Andreas Orthaber Department of Chemistry, Ångström Laboratory, Uppsala University, 75120 Uppsala, Sweden; orcid.org/0000-0001-5403-9902
- Daniel J. V. A. dos Santos Research Institute for Medicines (iMed.ULisboa), Faculty of Pharmacy, Universidade de Lisboa, 1649-003 Lisbon, Portugal; LAQV@REQUIMTE/ Department of Chemistry and Biochemistry, Faculty of Sciences, University of Porto, 4169-007 Porto, Portugal

Complete contact information is available at: https://pubs.acs.org/10.1021/acs.orglett.0c03647

Author Contributions

All authors have given approval to the final version of the manuscript.

Notes

The authors declare no competing financial interest.

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