



PROCEEDINGS OF THE
22nd International Symposium
on Analytical and Environmental Problems

October 10, 2016

**University of Szeged, Department of Inorganic and
Analytical Chemistry**



Szeged
Hungary

Edited by:
Tünde Alapi
István Ilisz

Publisher:
University of Szeged, Department of Inorganic and Analytical
Chemistry, H-6720 Szeged, Dóm tér 7, Hungary

ISBN 978-963-306-507-5

2016.
Szeged, Hungary

***The 22nd International Symposium
on Analytical and Environmental Problems***

Organized by:

SZAB Kémiai Szakbizottság Analitikai és Környezetvédelmi Munkabizottsága

Supporting Organizations

*University of Szeged, Department of Inorganic and Analytical Chemistry
Hungarian Academy of Sciences*

Symposium Chairman:

István Ilisz, PhD

Honorary Chairman:

Zoltán Galbács, PhD

Organizing Committee:

István Ilisz, PhD

associate professor

University of Szeged Department of Inorganic and Analytical Chemistry

ilisz@chem.u-szeged.hu

Tünde Alapi, PhD

assistant professor

University of Szeged Department of Inorganic and Analytical Chemistry

alapi@chem.u-szeged.hu

TiO₂ COATED MEMBRANE SURFACE CHARACTERISATION WITH CONTACT ANGLE MEASUREMENTS

Ildikó Kovács, Szabolcs Kertész, Gábor Veréb, Cecilia Hodúr, Zsuzsanna László*

*Department of Process Engineering, Faculty of Engineering, University of Szeged, H-6725 Szeged, Moszkvai krt. 9., Hungary
e-mail: zsizsu@mk.u-szeged.hu*

Abstract

In this work a neat polyacrylonitrile ultrafiltration membrane performance was compared to a same but TiO₂ coated membrane during oily water filtration. The fouling mechanisms of the two membranes and the resistances were determined and calculated. It was found that in case of the neat membrane initially intermediate pore blocking occurs that is followed by cake layer formation. The TiO₂ coating proved to be beneficial due to its fouling mitigating effect. With contact angle measurements the membrane wettability changes were monitored, which occurred as a result of the TiO₂ coating, oily water filtration and cleaning of the membrane surface by UV irradiation. After 3 hours of UV irradiation the water flux reached the neat membrane flux, but the residual oil layer on the surface still had a significant effect on the surface wettability.

Introduction

The rapid growth of oil and gas, petrochemical, pharmaceutical, metallurgical and food industries resulted in large oily wastewater production. Physical and chemical wastewater treatment methods are widely investigated, taking into consideration their efficiency, cost, need of additives, equipment and infrastructure, process time, capacity, etc. [4], but most of them are not effective enough to treat stable oil in water emulsions. Membrane filtration is an efficient process to treat oily wastewaters, without chemical additives and low energy cost compared to traditional separation methods [1, 2, 5]. Ultrafiltration is the most effective treatment, among membrane processes for this purpose [5]. Polymer membranes are the most commonly used type of membranes in water and wastewater treatment. Despite their beneficial qualities, their separation performance, antifouling property and long-time stability need improvement [1-4].

As the membrane fouling is a key problem; in order to minimize it membrane surfaces can be modified with photocatalytic nanoparticles. One way of modifying the membrane is to deposit the particles on the membrane surface [6–8]. TiO₂ is the most commonly used photocatalyst due to its good physical and chemical properties, availability, high photocatalytic activity and desirable hydrophilic properties [6–9]. Increasing the membrane hydrophilicity can increase the membrane fouling resistance [9]. Contact angle measurements are commonly used to determine the wettability of a surface. Contact angles also can show an accordance with the surface roughness [10].

The aim of this work was to investigate the effect of TiO₂ coating on a polyacrylonitrile ultrafiltration membrane performance during oily water filtration compared to the neat membrane. Furthermore, to characterize the membrane surface wettability changes by contact angle measurements after TiO₂ coating, oily water filtration, and a cleaning process which consisted of rinsing the membrane surface with water and then irradiating it for 1, 2 and 3 hours with UV light.

Experimental

The membrane was a polyacrylonitrile membrane (VSEP, New Logic Research Inc. USA) with 50kDa cut off weight, and membrane surface area 0.00342 m². Commercial TiO₂ (AEROXIDE P25, Evonik Industries) was used to coat the membrane surface. The membranes were coated with TiO₂ by filtering through the membrane 25mL, 0.4 g/L catalyst suspension in a dead end cell, at 1 MPa without stirring, that resulted in 0.3mg/cm² TiO₂ coating. The filtration was carried out with a Millipore batch filtration unit (XFUF04701, Solvent-resistant Stirred Ultrafiltration Cell, Millipore, USA). For the cleaning cycles with UV irradiation its cap was modified so that the UV light source can be fitted in it. This way a photocatalytic membrane reactor was set up. The UV light source was a low pressure mercury-vapor-lamp (GERMIPAK LightTech, Hungary, 40W, λ=254nm). The filtrations of the model wastewater were carried out at 0.3 MPa transmembrane pressure, without stirring at 20°C.

The model wastewater (oil in water emulsion) was prepared from crude oil (Algyő-area, Hungary) and distilled water. First a 1 wt.% emulsion was prepared by intensive stirring (35000 rpm), then 5 mL of this emulsion was inoculated into 495 mL of distilled water directly below to the transducer of an ultrasonic homogenizer (Hielscher UP200S) resulting the oil in water emulsion (coil=100 ppm). Time of homogenization was 10 minutes, maximal amplitude and cycle was applied while the emulsion was thermostated to 25°C. In case of every filtration 250 mL oily water portion was filtered to volume reduction ratio (VRR) 5. VRR [-], was defined as:

$$\text{VRR} = \text{VF} / (\text{VF} - \text{VP}) \quad (1)$$

where VF and VP is the volume of the feed and permeate [m³] respectively at any time.

Hermia's model was applied to determine the fouling mechanism of each membrane during filtration [11,12]. Resistance-in-series model was applied to analyze resistances that lead to flux decline during the ultrafiltration process[11].

Membrane hydrophobicity was quantified by measuring the contact angle that was formed between the (neat, TiO₂ coated and oily contaminated) membrane surface and distilled water. 10µL water was carefully dropped on the top of the membrane surface. Contact angles were measured using the sessile drop method (Dataphysics Contact Angle System OCA15Pro, Germany). In every case the membrane was sliced up and the entire surface was characterized.

Results and discussion

In the first series of experiments the difference between the fouling properties of the neat and TiO₂ coated membranes were investigated. The flux decline in case of the TiO₂ coated membrane was significantly lower compared to the neat membrane (Fig. 1). In the initial stage of the filtration the intermediate pore blocking describes the best the measured data which is followed by the cake layer formation (Fig. 1.a). In case of the TiO₂ coated membrane all blocking models fit the measured data similarly (Fig. 1.b), for that the explanation lies in the resistances. The calculated resistances give additional information about the mechanism of the fouling. Filtration of oily water through the neat membrane results in a high reversible resistance and irreversible resistance, compared to the TiO₂ coated membrane (Fig. 2), meaning that first complete pore blocking occurs and that is followed by the oil cake layer forms on the membrane surface which can be removed by rinsing the membrane with water. The reversible and irreversible resistances in case of the TiO₂ coated membrane are lower than the membrane resistance, which makes them marginal and explains why the blocking models fit the measured data so closely because none of them occurs to a significant extent (Fig. 1.b and 2).

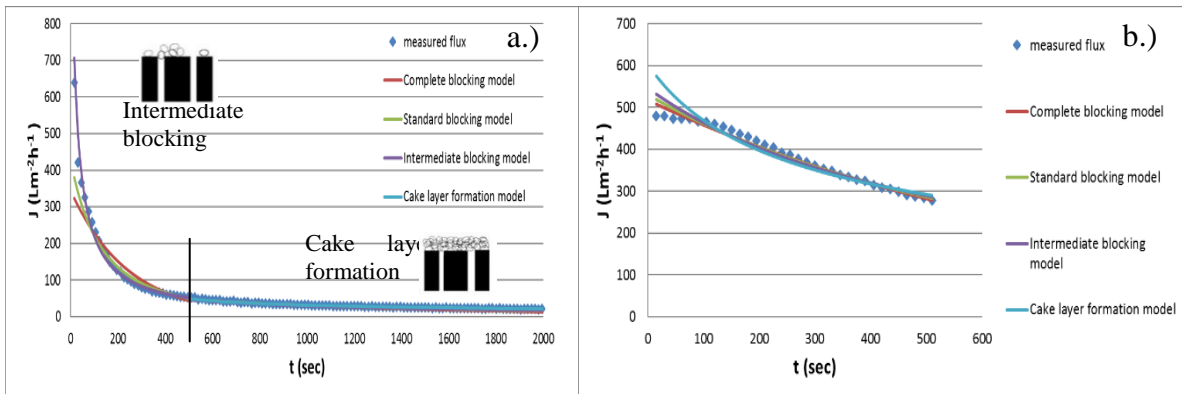


Figure 1. Flux decline and fitted fouling models during oily water filtration through the neat (a) and TiO₂ coated (b) PAN membranes

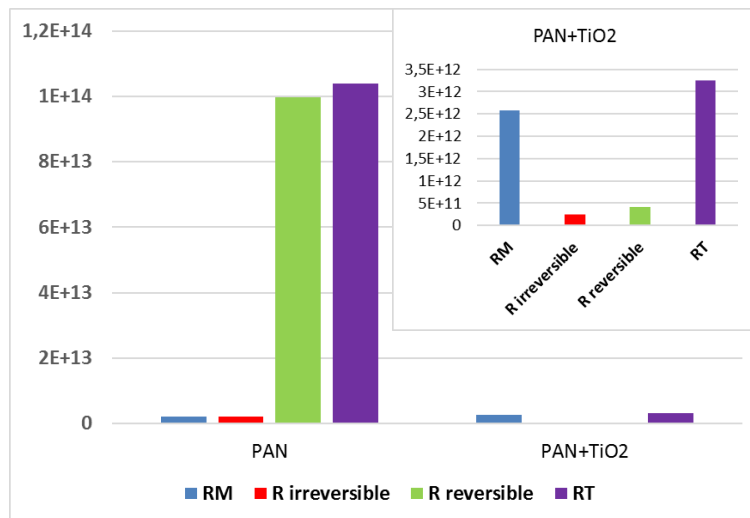


Figure 2. Resistances in case of oily water filtration through the TiO₂ coated and neat PAN membranes

In the second series of experiments the wettability changes of the membrane during coating, oily water filtration, and UV cleaning were measured to determine if the cleaning process of the surface can be monitored with contact angle measurements. By coating the membrane with TiO₂ the surface wettability increases (Fig. 3), the catalyst forms a hydrophilic layer on the membrane surface.

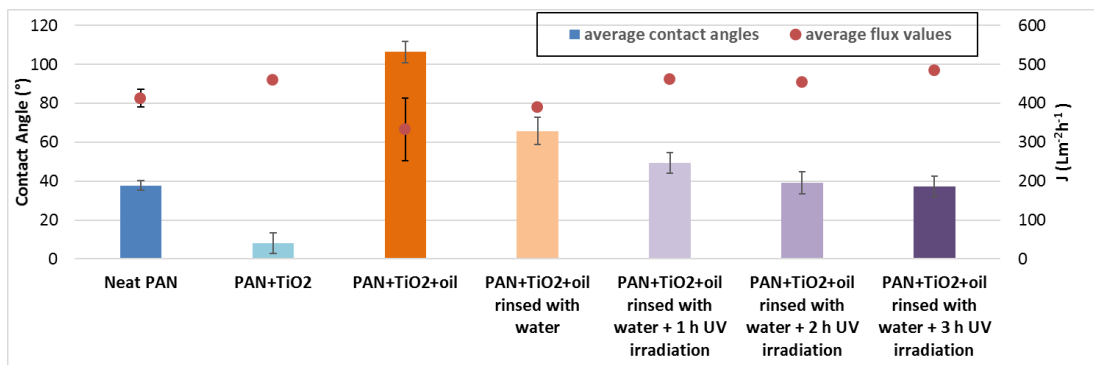


Figure 3. Average contact angles and average flux values of the neat, TiO₂ coated and then fouled and cleaned PAN membranes

The TiO₂ layer does not change the water flux significantly (Fig. 3). During the filtration of

the oily water a hydrophobic oil layer forms on the surface, which results in the decrease of the flux (Fig. 3). By rinsing the membrane surface with water the oil layer can be removed to an extent that results in a flux and hydrophilicity increase (Fig. 3). The wider error bar on the flux of TiO₂ coated membrane during oily water filtration represents the extent of fouling during filtration. The UV irradiation of the membrane surface with time results in higher flux and hydrophilicity (Fig. 3). After 3 hours the water flux reached the neat membrane flux (Fig. 3). The average hydrophilicity of the membrane after 3 hour is nearly the same as the neat membrane hydrophilicity, but to recover the hydrophilicity of the TiO₂ coated membrane a longer irradiation time is needed. The hydrophilicity increase indicates the degree of the decomposition of the oil layer.

Conclusion

During oily water filtration on the neat polyacrylonitrile membrane initially intermediate blocking occurs that is followed by cake layer formation, that results in a rapid flux decline. This means that the oil layer acts as an additional filtration layer. By coating the membrane surface with TiO₂, the membrane becomes more resistant to fouling, and does not come to oil cake layer formation. The flux decline during oily water filtration is moderate compared to the neat membrane. The reversible and irreversible resistances are negligible compared to the membrane resistance. The oil that is on the TiO₂ coated membrane surface after filtration significantly increases the membrane surface hydrophobicity that can be reduced by rinsing the membrane surface with water, which results in significant flux recovery. By UV irradiation the surface hydrophilicity can be further increased and total flux recovery can be achieved within 3 hours.

Acknowledgements

This project was supported by the János Bolyai Research Scholarship of the Hungarian Academy of Sciences. The authors are also grateful for the financial support provided by the project Hungarian Science and Research Foundation (NKFI contract number K112096).

References

- [1] X.S. Yi, S.L. Yu, W.X. Shi, N. Sun, L.M. Jin, S. Wang, B. Zhang, C. Ma, L.P. Sun, *Desal.*, 281 (2011) 179–184.
- [2] B. Hu, K. Scott, *Chem. Eng. J.*, 136 (2008) 210–220.
- [3] S. Leong, A. Razmjou, K. Wang, K. Hapgood, X. Zhang, H., *J. Mem. Sci.*, 472 (2014) 167–184.
- [4] M. Padaki, R. S. Murali, M.S. Abdullah, N. Misdan, A. Moslehyani, M.A. Kassim, N. Hilal, A.f. Ismail, *Desal.*, 357 (2015) 197–207.
- [5] Y. He, Z. W. Jiang, *Filt. Sep.*, 45(5) (2008) 14–16.
- [6] H. Bai, Z. Liu, D. D. Sun, *Chem. Commun.* 46 (2010) 6542–6544.
- [7] R. Molinari, L. Palmisano, E. Drioli, M. Schiavello, *J. Membrane Sci.* 206 (2002) 399–415.
- [8] E. Bet-moushoul, Y. Mansourpanah, Kh. Farhadi, M. Tabatabaei, *Chem. Eng. J.* 283 (2016) 29–46.
- [9] Y. Mansourpanah, S.S. Madaeni, A. Rahimpour, A. Farhadian, A.H. Taheri, *J. Membrane Sci.* 330 (1–2) (2009) 297–306.
- [10] K.C. Khulbe, C. Feng, T. Matsuura, G.C. Kapantaidakis, M. Wessling, G.H. Koops, *J. Membrane Sci.* 226 (2003) 63–73.
- [11] M. C. V. Vela, S. A. Blanco, J. L. García, E. B. Rodríguez, *Sep. Purif. Technol.* 62 (2008) 489–498.
- [12] J. Hermia, *Trans. Inst. Chem. Eng.* 60 (1982) 183–187.